

3. Which oxides cause acid rain ? What is its pH value ?
4. The empirical formula of a compound is CH_2O . Its molecular weight is 90. Calculate the molecular formula of the compound.
5. What is Chemical Oxygen Demand (COD) ?
6. How does graphite function as a lubricant ?
7. How is water gas prepared ?
8. State the first law of the thermodynamics.
9. What is the relationship between C_p and C_v ?
10. Write IUPAC names of the following compounds :
 - (i) $\text{CH}_2 = \underset{\text{CH}_3}{\text{C}} - \text{CH}_3$
 - (ii) $\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH}_2$

SECTION B

6×4=24

Note :—Answer ANY SIX questions.

11. Deduce :
 - (a) Boyle's law, and
 - (b) Charles's law
 from kinetic gas equation.
12. Explain the hybridization involved in PCl_5 molecule.
13. Explain the formation of coordinate covalent bond with one example.
14. Balance the following redox reaction by Ion-electron method :

$$\text{MnO}_4^- + \text{SO}_2 \rightarrow \text{Mn}^{2+} + \text{HSO}_4^-$$

(aq) (g) (aq) (aq)

(acidic solution)
15. Discuss the application of Le-Chatelier's principle for the synthesis of Ammonia by Haber's process.
16. What is hard water ? Write a note on the Calgon's method for the removal of hardness of water.

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