



Student Area

This assignment is due at 11:45 PM on 05/25/2021. Any work submitted afternoon that was past 11:45 PM on 05/25/2021 will result in an automatic 10 point deduction. Also, any work submitted between 11:45 PM on 05/25/2021 and 11:45 PM on 05/25/2021 will result in an additional 10 points being deducted for the assignment.

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Due date passed.

Check back several days after the due date to view points added by your instructor.

Attendance Points = 5

Experiment 3 Report Sheet

SYNTHESIS OF AN ALUM $KAl(SO_4)_2 \cdot 12H_2O$

Total Points =

Current	Goal
0/1	1/0

Introduction

Introduction

Note: All essay answers throughout the semester must be composed in the **blue text editor** below.

1/0 1/0

The objectives of experiment 3 were to synthesize the alum $KAl(SO_4)_2 \cdot 12H_2O$, separate the product from the reaction mixture, purify the product, and assess its purity. The described objectives were accomplished by synthesizing $KAl(SO_4)_2 \cdot 12H_2O$ from Al, KOH, and H_2SO_4 . After the reactions occurred, the $KAl(SO_4)_2 \cdot 12H_2O$ was precipitated by cooling the product. The precipitated product was then washed with ice water to purify it by

Chem 21 Lab Answers

Scott C. Dulebohn



Chem 21 Lab Answers:

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