

Name: KEY Date: \_\_\_\_\_ Period: \_\_\_\_\_

### Radioactivity Worksheet

1. State the number of neutrons and protons in each of the following nuclei:

- a.  ${}^3_1\text{H}$  : Neutrons = 1    protons = 1  
b.  ${}^{12}_6\text{C}$  : neutrons = 6    protons = 6  
c.  ${}^{56}_{26}\text{Fe}$  : neutrons = 30    protons = 26  
d.  ${}^{197}_{79}\text{Au}$  : neutrons = 118    protons = 79

2. The three types of radioactive emissions are called alpha ( $\alpha$ ), beta ( $\beta$ ) and gamma ( $\gamma$ ) radiation. Complete the table below with the correct information about each type.

	Charge	Atomic Symbol	Can Be Stopped By
Alpha	+2	${}^4_2\text{He}$	paper
Beta	-1	${}^0_{-1}\text{e}$	aluminum
Gamma	0	${}^0_0\gamma$	lead

3. Which of the three radioactive emissions ( $\alpha$ ,  $\beta$ ,  $\gamma$ ) best fit the following statements? Write the correct symbol/s on the lines.

- a) These emissions are charged.  $\alpha, \beta$   
b) This emission is the most massive (heaviest).  $\alpha$   
c) This emission is the most charged.  $\alpha$   
d) This emission is most dangerous outside of the body.  $\gamma$   
e) This emission is stopped by thin paper or a few centimeters of air.  $\alpha$   
f) This emission can travel through paper, but is stopped by aluminum.  $\beta$   
g) This emission can travel through fairly thick lead.  $\gamma$

4. Which type of radiation – alpha, beta, or gamma:

- a. Results in the greatest change in atomic number? Why?

Alpha - changes by 2

- b. Results in the least change in atomic number? Why?

Gamma - no change in atomic number

# Chapter 9 Section 1 Radioactivity Worksheet Answers

**LM Reder**



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