

## Chapter Test A

### Chapter: Chemical Equations and Reactions

Do the work provided below the letter of the item or place the item number next to the statement or brief answer each question.

1. Which statement is true?
  - a. The number of atoms is conserved.
  - b. The number of molecules is not conserved.
  - c. The number of molecules is not conserved, but the number of atoms is conserved.
  - d. None of the above; all three items listed are false statements.
2. In the reaction described by a chemical equation,
 
$$\text{reactants} \rightarrow \text{products} + \text{products} + \text{products} + \text{products}$$
  - a. all the reactants are products.
  - b. all the reactants are products.
  - c. all the reactants are products.
  - d. None is true.
3. Which of the following relationships is always true?
  - a. The products of a reaction are always different from the reactants of the reaction.
  - b. All of the reactants are products.
  - c. The number of molecules is conserved in all reactions.
  - d. The number of atoms is conserved in all reactions.
4. Which is a correct statement?
  - a. The number of atoms is conserved in all reactions.
  - b. The number of molecules is conserved in all reactions.
  - c. The number of atoms is conserved in all reactions.
  - d. All of the above.
5. In a reaction, a chemical equation tells you
  - a. what the reactants are.
  - b. what the products are.
  - c. what the reactants and products are.
  - d. what the coefficients are.

# Chapter 9 Modern Chemistry Test

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