

Atoms, Ions, & Molecules

	Atom	Ion	Molecule
Number of particles	 One 	One or more   	Two or more 
Charge	Neutral (equal protons and electrons)	Positive (lost electrons) negative (gained electrons)	Neutral (electrons shared between atoms)
Formation	Cannot be further broken down without losing chemical identity.	Formed by losing or gaining electrons.	Formed by sharing electrons between atoms.
Examples	Helium (He), Oxygen (O), Iron (Fe)	Sodium ion (Na ⁺), Chloride ion (Cl ⁻), Sulfate ion (SO ₄ ²⁻)	Water (H ₂ O), Carbon dioxide (CO ₂), Sugar (C ₆ H ₁₂ O ₆)



Chapter 7 Compounds Ions And Molecules

**Robert M. Silverstein,G. Clayton
Bassler, Terence C. Morrill**



Chapter 7 Compounds Ions And Molecules:

Chapter 7 Compounds Ions And Molecules Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 7 Compounds Ions And Molecules**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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