

CHAPTER 14 NUCLEI

Nuclear Chemistry

SECTION 1

Write a suitable answer for the following questions in the space provided.

1. Describe the transition of the three primary particles in the case of beta minus decay.
2. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) Magnesium
 - b) Aluminium
 - c) Silicon
 - d) Phosphorus
3. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) in atomic number 38
 - b) in atomic number 32
 - c) in atomic number 30
 - d) in atomic number 34
4. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
5. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
6. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
7. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
8. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
9. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
10. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
11. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
12. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
13. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
14. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
15. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
16. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
17. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
18. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
19. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons
20. The number of neutrons in the nucleus of an element is 20. The number of protons is 18. The element is:
 - a) number of neutrons
 - b) number of protons
 - c) number of electrons
 - d) ratio of neutrons to protons

Chapter 22 Review Nuclear Chemistry Answer Key

U.S. Nuclear Regulatory Commission



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