

SOME BASIC CONCEPT OF CHEMISTRY

FORMULA - SHEET

BY - JAY SIR (J*)

① $PV = nRT$

P = Pressure (atm)
V = Volume (Litre)
n = mole
R = gas constant (0.0821 atm L/mol K)
T = Temperature (Kelvin 'K')

② $d_{\text{gas}} = \frac{PM}{RT}$

M = Molar mass
Baki salt upar hai

③ $\text{Density} = \frac{\text{mass}}{\text{volume}}$

④ $\frac{F - 32}{9} = \frac{C}{5} = \frac{K - 273}{5}$

F = Degree Fahrenheit
C = Degree Centigrade
K = Degree Kelvin

⑤ $1\text{m}^3 = 1000\text{L}$
 $1\text{dm}^3 = 1\text{L}$
 $1\text{cm}^3 = 1\text{ml}$

⑥ $1\text{amu} = \frac{1}{12} \times \text{mass of one } ^{12}\text{C isotope}$

$1\text{amu} = 1.66 \times 10^{-24}\text{gm}$

$1\text{amu} = \frac{1}{N_A}\text{gm}$

⑦ $\text{mole} = \frac{\text{mass (gm)}}{\text{molar mass}}$

$\text{mass} = \text{mole} \times \text{molar mass}$

⑧ $\text{mole} = \frac{\text{no. of particles (atom/molecule)}}{6.022 \times 10^{23}}$

$\text{no. of particles} = \text{mol} \times N_A$

⑨ $\text{mole} = \frac{\text{volume of gas at NTP/STP}}{22.4}$

$V_{\text{gas}} = \text{mol} \times 22.4$

⑩ मोल मात्र

"Kuch bhi diya ho, sabse pehle mole nikalo fir jo puchhe, wo mole ki help se nikalo"

$\text{mole} = \frac{W}{M} = \frac{N}{N_A} = \frac{V_L}{22.4}$

Assessment Of Basic Chemistry Concepts Answer Sheet

G Thomas



Assessment Of Basic Chemistry Concepts Answer Sheet:

Assessment Of Basic Chemistry Concepts Answer Sheet Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Assessment Of Basic Chemistry Concepts Answer Sheet**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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