



Separate Chemistry

What is a substance named?

What are examples of these substances named?

What are the following named as one of metals, a solution or none of the options?

What is this in a big box

--	--	--	--	--	--	--	--	--	--

Why are hydrogen and water substances named as the mixture?

What is named as the mixture when other hydrogen and water mixing?

Why is sulphur found in its pure state?

Complete the acid reactions

acid + carbonate → carbon dioxide +

acid + hydroxide → hydrogen oxide +

acid + metal → salt +

acid + hydrogen → hydrogen salt +

Describe what a metal reacting with an acid can tell you about the reactivity of the metal.



What are the products?

- iron + acid

What are the products?



How much rain will water to produce

How much rain will acid to produce

Is named as you can see. (solid food)

- orange + acid
- orange + acid
- orange
- orange + acid
- orange + acid
- orange

Describe how to make a soluble salt from an insoluble base

1. Choose an acid
2. Choose an insoluble base
3. Weigh the acid
4. Add the insoluble base to the acid until there is no further reaction
5. Filter the mixture
6. Wash the residue with distilled water
7. Evaporate the water to dry

Complete the acid reactions



What is the pH of the products of a neutralisation reaction?

7 = 7 = 7 = 7

Complete the following

- 1. _____
- 2. _____
- 3. _____
- 4. _____
- 5. _____
- 6. _____

_____ is the acid formed and _____ is the salt formed

Worksheet for 'Separate Chemistry' with a grid for notes and a diagram of a titration setup. The diagram shows a burette containing a blue liquid being added to a conical flask containing a yellow liquid. Labels include 'Burette', 'Conical flask', and 'Indicator'. The grid contains various text boxes and lines for student input.

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