

AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which pair of atoms should form the most polar bond?
a. F and B
b. C and O
c. F and O
d. N and F
e. B and N
- _____ 2. Which pair of ions should form the ionic lattice with the highest energy?
a. Na^{1+} and Br^{1-}
b. Li^{1+} and F^{1-}
c. Cs^{1+} and F^{1-}
d. Li^{1+} and O^{2-}
e. K^{1+} and F^{1-}
- _____ 3. The electrostatic force of attraction is greatest in which compound?
a. BaO
b. MgO
c. CaS
d. MgS
e. CaO
- _____ 4. Which molecule has the weakest bond?
a. CO
b. O_2
c. NO
d. N_2
e. Cl_2
- _____ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?
a. BCl_3
b. SO_2
c. CO_2
d. BeF_2
e. SO_4^{2-}
- _____ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?
a. $\text{O}-\text{O}-\text{O}$
b. $\text{O}=\text{O}-\text{O}$
c. $\text{O}=\text{O}=\text{O}$
d. $\text{O}=\text{O}=\text{O}$
e. $\text{O}=\text{O}=\text{O}$
- _____ 7. Which species has the shortest bond length?
a. CN^{1-}
b. O_2
c. SO_2
d. SO_3
e. CO_2
- _____ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?
a. Na^{1+}
b. He
c. F^{1-}
d. O^{2-}
e. Mg^{2+}
- _____ 9. Which species has a valid non-octet Lewis structure?
a. GeCl_4
b. SiF_4
c. CO_3^{2-}
d. SeCl_4
e. CO_3^{2-}

Ap Chemistry Chapter 9 Test Zumdahl

Laura L. Duncan



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Study Guide for Chemistry by Steven S. Zumdahl Martha B. Barrett, 1986 **Chemistry** Steven S. Zumdahl, 1993

Introductory Chemistry Steven S. Zumdahl, 1996 **Basic Chemistry** Steven S Zumdahl, 1996 *CliffsNotes AP*

Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **AP Chemistry** Albert S.

Tarendash, Frederick J. Rowe, 2002 This indispensable guide to chemistry helps students who wish to prepare for the AP Chemistry exam on their own Comprehensive and easy to understand this learning guide includes a full content review two full length practice tests with hundreds of practice questions and thorough answer explanations and proven test taking strategies *Barron's AP Chemistry* Neil D. Jespersen, 2010-02-01 Updated to reflect the most recent Advanced Placement

Chemistry exams this manual presents three diagnostic tests and three full length practice exams all with questions answered and explained The author also presents additional routine and more challenging problems in every chapter A complete review of AP chemistry topics covers the structure of matter chemical bonding states of matter physical chemistry chemical reactions and other topics that appear on the AP exam *Cracking the AP Chemistry Exam, 2018 Edition* Princeton Review, 2017-08 Provides subject review for the AP chemistry exam and includes two full length practice tests

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[Cracking the AP Chemistry Exam, 2014 Edition \(Revised\)](#) Princeton Review,2014-03-18 THE PRINCETON REVIEW GETS RESULTS Get all the prep you need to ace the revised AP Chemistry Exam with 2 full length practice tests thorough topic reviews and proven techniques to help you score higher The AP Chemistry course and test are undergoing major changes with a new version of the exam debuting in May 2014 Inside Cracking the AP Chemistry Exam you ll find 2 full length practice tests with detailed explanations that include the new multiple choice and constructed response question types Expert subject reviews for all test topics that reflect the changes to the 2014 AP Chemistry exam including newly incorporated test topics and Big Ideas organization Practice drills at the end of each chapter Step by step strategies techniques for every section of the exam A comprehensive list of key chemistry equations and constants This eBook edition has been specially formatted for on device viewing with cross linked questions answers and explanations **AP Chemistry Premium, 2022-2023: Comprehensive Review with 6 Practice Tests + an Online Timed Test Option** Neil D.

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