

Chemistry – Unit 4 Notes Dalton's Playbook

In the late 19th century, British physicist John Dalton conducted and refined an atomic model which explained experimental observations. Dalton's theory of atoms and molecules explained several facts. The substance in which atoms combine to form molecules is called a molecular solid. Such solids will retain their composition and chemical properties throughout a change in temperature under suitable conditions of pressure and temperature.

Table 1 – Properties

State	1800s	1850s	1900s
Atomic model proposed			
Properties explained			

1. What happened to the mass of the particles in the models as it was refined?
2. What changes were observed in terms of the composition and the physical properties in the models?
3. What are the similarities between the solid and gas particles and the mass of the particles was found.

Table 2 – Examples

This table could be used to record the actual and theoretical properties of particles in each of the three states: solid, liquid and gas.

	Solid state	Liquid state	Gaseous state
Atomic model proposed			
Atomic model refined			
Atomic model proposed			
Properties explained			
Atomic model refined			

1. What evidence for the existence of the particles in different states (solid, liquid and gas) were found?
2. How did the mass of the particles in different states (solid, liquid and gas) compare to the mass of the particles?

Modeling Chemistry Dalton Playhouse Notes Answers

Judd E. Hollander



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The Journal of Education ,1931 **The Scientific Traveler** Charles Tanford, Jacqueline Ann Reynolds, 1992 “The”
Illustrated London News ,1850 **New York Magazine** ,1997-04-28 New York magazine was born in 1968 after a run as
an insert of the New York Herald Tribune and quickly made a place for itself as the trusted resource for readers across the
country With award winning writing and photography covering everything from politics and food to theater and fashion the
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