

## Ideal gas law - problems and solutions

1. Ideal gases in a closed container initially have volume  $V$  and temperature  $T$ . The final temperature is  $5/4T$  and the final pressure is  $2P$ . What is the final volume of gases.

Known :

Initial volume ( $V_1$ ) =  $V$

Initial temperature ( $T_1$ ) =  $T$

Final temperature ( $T_2$ ) =  $5/4 T$

Initial pressure ( $P_1$ ) =  $P$

Final pressure ( $P_2$ ) =  $2P$

Wanted : Final volume ( $V_2$ )

Solution :

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$\frac{P V}{T} = \frac{(2P) V_2}{5/4 T}$$

$$\frac{V}{1} = \frac{2 V_2}{5/4}$$

$$5/4 V = 2 V_2$$

$$V_2 = \frac{5V}{4} \cdot \frac{2}{1}$$

$$V_2 = \frac{5V}{4} \times \frac{1}{2}$$

$$V_2 = \frac{5V}{8}$$

2. Determine the volume of 2.00 moles of gases (ideal gas) at STP. STP = Standard Temperature and Pressure.

Known :

Moles of gas ( $n$ ) = 2 moles

Standard temperature ( $T$ ) =  $0^\circ\text{C} = 0 + 273 = 273$  Kelvin

Standard pressure ( $P$ ) = 1 atm =  $1.013 \times 10^5$  Pa

Universal gas constant ( $R$ ) = 8.315 Joule/mole.Kelvin

Wanted : Volume of gases ( $V$ )

Solution :

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<https://physics.gurumuda.net/ideal-gas-law-problems-and-solutions.htm>

# Gas Law Problems With Solutions

**Minjie Lin**



## Gas Law Problems With Solutions:

**Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields *Dr. Salm's Chemistry Problem Drill Book* Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests Math Concepts for Food Engineering Richard W. Hartel, D.B. Hyslop, Terry Howell, Jr., 1997-05-19 Food industry managers engineers and scientists require a personal repertory of mathematical tools to find solutions to a variety of work problems These problems relate to processing research analysis and other operations that require mathematical solutions This new handbook provides a practical review of required basic mathematical methods and relates them to specific areas of food industry problem solving Prepared for students in undergraduate food science programs the handbook also serves the needs of food industry technical personnel who would be helped by a review of basic math for food engineering and non technical managers who seek an introduction to food engineering math and problem solving Each chapter on a math or problem solving skill begins with fundamental concepts presents step by step procedures in the skill and provides numerous practical illustrative examples Practice problems with answers in the appendix provide for self testing and reinforcement of learning All of the math skills presented relate to common food industry problems From the text In food engineering and any other mathematically oriented discipline one must become adept at problem solving using equations and mathematical principles Many of the problems that food scientists and engineers face require the ability to apply physical and mathematical principles to solve equations and calculate necessary information Thus problem solving abilities are extremely important and we should be familiar with the general steps in problem solving Journal of Chemical Education, 1925

Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Questions and Answers on Federal Tax Laws, Problems Affecting Individuals, Partnerships and Corporations, January, 1924 Irving Bank-Columbia Trust Company,1924 Physico-chemical Calculations Joseph Knox,1919 *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **University High School Journal** ,1925 *Student Solutions Manual for Chang's Chemistry* Philip C. Keller,Raymond Chang,Jill L. Keller,1988 **Journal** University High School (Oakland, Calif.),1926 *Physico-chemical Calculations* Joseph Knox,1944 Chemical Calculations Raymond Harman Ashley,1929 **eBook: General, Organic and Biological Chemistry 2e** SMITH,2012-02-16 eBook General Organic and Biological Chemistry 2e Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Journal of the American Chemical Society** American Chemical Society,1921 Proceedings of the Society are included in v 1 59 1879 1937 **Chemical News and Journal of Industrial Science** ,1908 *The Chemical News and Journal of Physical Science* ,1913 *The Journal of Physical Chemistry* Wilder Dwight Bancroft,1912 Includes section New Books *The Journal of Physical Chemistry* ,1912 *The Chemical News and Journal of Industrial Science* William Crookes,James H. Gardiner,Gerald Druce,H. W. Blood-Ryan,1908

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