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Fundamental Applied Maths Solutions

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 $2c_2h_2 + 5o_2 \rightarrow 4co_2 + 2h_2o$
1 2 0 mol if 1 0 mol of c_2h_2 is burned how many moles of co_2 are formed 2 5 mol c if 1 0 mol of c_2h_2 is burned how many moles of oxygen gas are consumed 9 a

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