

USMLE STEP 1 BIOCHEMISTRY EXAM 2025/2026 QUESTIONS WITH ANSWERS GRADED A+

✓✓What type of Antibiotics inhibit bacterial topoisomerase 4? - ✓✓Quinolones

✓✓_____chromatin is tightly coiled and transcriptionally inactive while _____chromatin is less condensed and is transcriptionally active. - ✓✓Heterochromatin; Euchromatin

✓✓DNA methylation often _____ genes while histone acetylation _____ -

✓✓Inactivates (decreases gene transcription); Loosens DNA making it more accessible and more easily transcribed.

✓✓Sjogren's syndrome (lymphocytic infiltration of salivary and lacrimal glands resulting in dry mouth/dryeyes) is often associated with antibodies against SS-A/B (single strand binding proteins). What role do these proteins play in prokaryotic DNA replication? - ✓✓SS-DNA binding protein binds and stabilizes ssDNA to prevent re-annealing

✓✓DNA polymerase III has polymerase activity in the _____ direction and exonuclease activity in the _____ direction. - ✓✓5 to 3; 3 to 5

✓✓The primer for DNA replication synthesized by primase is made of DNA or RNA? - ✓✓RNA

✓✓RNA primer removal in DNA synthesis is done by DNA polymerase _____. - ✓✓1

✓✓What DNA replication enzyme is responsible for sealing the remaining nicks? - ✓✓DNA ligase

✓✓What are the 3 stop codons? - ✓✓UGA, UAG, UAA

✓✓Proteins are synthesized from ____ to ____ terminus. - ✓✓N; C

✓✓1 codon for ____ amino acid. - ✓✓1

✓✓DNA to RNA is known as _____. - ✓✓Transcription

✓✓RNA to protein is known as _____. - ✓✓Translation

✓✓What is the Translation start codon and what amino acid does it code for? - ✓✓AUG; Methionine

Active Biochemistry Answers

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transformations can occur as a result of nuclear decay of radioactive isotopes or neutron capture neither procedure requires a change in the atom s element an atom gets transmuted into an atom of a different element when the number of protons in it is changed

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nuclear chemistry is the subfield of chemistry dealing with radioactivity nuclear processes such as nuclear transmutation and nuclear properties it is the chemistry of radioactive elements such as the actinides radium and radon together with the chemistry associated with equipment such as nuclear reactors which are designed to perform

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unlike a normal chemical reaction that forms molecules nuclear reactions result in the transformation of one element into another this property of nuclear reactions is used in nuclear power plants to gather nuclear energy the three common types of radiations are explained below

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a nuclear reactor consists of the following a nuclear fuel a fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction the radioactive isotope is contained in tubes called fuel rods a moderator

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apr 3 2019 in order to understand the origin and character of individual radioactive emissions accompanying nuclear

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transmutation the changing of one chemical element into another transmutation involves a change in the nucleus or core of an atom and is therefore a nuclear reaction transformations may result from nuclear decay of radioactive isotopes or they can result from neutron capture

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in all accelerators the particles move in a vacuum to avoid collisions with gas molecules when neutrons are required for transmutation reactions they are usually obtained from radioactive decay reactions or from various nuclear reactions occurring in nuclear reactors

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nuclear reaction a nuclear reaction is a process in which two nuclei or a nucleus and an external subatomic particle collide to produce one or more new nuclides the nuclear reaction is a reaction that involves nucleoids the reactant nucleoid called the parent nucleoid usually transforms into a different nucleoid called the daughter

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nuclear transmutation is the conversion of one chemical element or an isotope into another chemical element 1 nuclear

transmutation occurs in any process where the number of protons or neutrons in the nucleus of an atom is changed a transmutation can be achieved either by nuclear reactions in which an outside particle reacts with a nucleus

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