

# ACIDS vs. BASES

On a **scale from 0 to 14**, lower pH solutions are **strong acids**; higher pH solutions are **bases**. Solutions measuring pH 7 are **neutral**.



**Acid** is a chemical substance that has a pH lower than seven.

An **acid** is any hydrogen-containing substance that is capable of donating a proton to another substance.

An **example of acid** substance can be found in the lemon juice which has a pH 2 level.



A **base** is a chemical substance that has a pH higher than seven.

A **base** is a molecule or ion able to accept a hydrogen ion from an acid.

An **example of base** substance can be found in the bleach which has a pH 12.6 level.

# Acids Bases And Ph

**Wolfgang Guggemos**



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