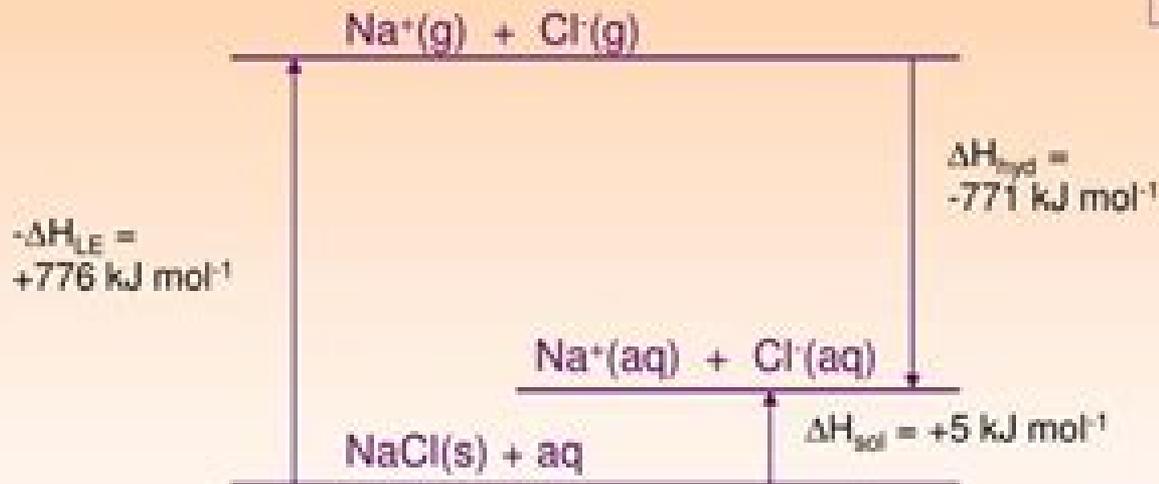


Enthalpy of Solution

- Enthalpy Diagram

Click to complete the diagram.



$$\Delta H_{\text{sol}} = -\Delta H_{LE} + (\Delta H_{\text{hyd}}(\text{cation}) + \Delta H_{\text{hyd}}(\text{anion}))$$

$$\Delta H_{\text{sol}} = -776 + -771 = +5 \text{ kJ mol}^{-1}$$

Enthalpy Of Solution Cacl2

David Kirk



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