
ACS CHEMISTRY FINAL EXAM STUDY GUIDE

- milli (m) (m has 3 lines, upside down it's a w (win) so negative=loss): 10^{-3}
- micro (inappropriate joke, i don't want Quizlet to ban me lol): 10^{-6}
- nano: 10^{-9}
- pico (p): 10^{-12}
- mono: one, single, alone
- di: two
- tri: three (tricycle)
- tetra: four (tetrapod) - dogs, cats, vertebrae animals
- penta: five (pentagon)
- hexa: 6 (hexagon)

- hepta: 7 (heptagon)
- octa: 8 (octopus, stop)
- nona: 9 (no = neign in German)
- deca: 10 (decade = 10 years)
- molar mass: the mass of one mole of a pure substance
- molar ratio: ratio of coefficients in a chemical equation
- limiting reactant: A reactant that is totally consumed during a chemical reaction, limits the extent of the reaction, and determines the amount of product.
- percent composition: the percent by mass of each element in a compound
- molarity: the number of moles of solute per liter of solution
- percent yield: $\text{actual yield/theoretical yield} \times 100$
- empirical formula: a chemical formula showing the ratio of elements in a compound rather than the total number of atoms
- molecular formula: A chemical formula that shows the number and kinds of atoms in a molecule, but not the arrangement of the atoms.
- gas laws: describe how the behavior of gases is affected by pressure, volume, and temperature

Chemistry Final Exam Study Guide

ML Morrison



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Study Guide for Chemistry by Steven S. Zumdahl Martha B. Barrett,1986

CliffsNotes AP Chemistry Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

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Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

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Florida Schools ,1964

Guide to Science Teaching in Secondary Schools Archie L. Lacey, Virginia M. Rowley,1966

Florida School Bulletin ,1963

Chemistry for the IB Diploma Geoff Neuss,2001 This concise guide provides the content needed for the Chemistry IB diploma at both Standard and Higher Level It follows the

structure of the IB Programme exactly and includes all the options Each topic is presented on its own page for clarity Higher Level material is clearly indicated and there are plenty of practice questions The text is written with an awareness that English might not be the reader's first language

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