

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ a. $\text{CH}_3\text{CO}^+\text{Cl}^-$
- _____ b. Lewis acid
- _____ c. Carbonyl chloride
- _____ d. AlCl_3
- _____ e. Boron trifluoride
- _____ f. Methyl cation
- _____ g. Methyl anion
- _____ h. Lewis base
- _____ i. Hydrogen ion
- _____ j. H^+

- _____ k. formed by sequential proton changes that are temperature or pH dependent
- _____ l. strong group of ionizable amino acids
- _____ m. the amino terminal amino acid that often forms crystalline compounds (strychnine, etc)
- _____ n. amino or imino bonds
- _____ o. noncovalent bonds are part of structure in chemical communication
- _____ p. the energy required to break a chemical bond and form product molecules
- _____ q. structural group of amino acids that hold together by hydrogen bonds, results from the drying of amino acid side chains
- _____ r. chemical bonding that results from the electrical attraction between positive charges
- _____ s. mutual electrical attraction between oppositely charged ions, resulting from electrostatic attraction, particularly the ionic region

Multiple Choice

2. The charge on a bond is _____

- _____ a. always positive
- _____ b. always negative
- _____ c. either positive or negative
- _____ d. neither

3. In many compounds, atoms of one group form bonds with atoms of another of another of the same compound, forming bonds of their own.

- _____ a. True
- _____ b. False
- _____ c. Yes
- _____ d. No

4. Which of the following is not a Lewis acid? _____

- _____ a. $\text{CH}_3\text{CO}^+\text{Cl}^-$
- _____ b. Lewis acid
- _____ c. Carbonyl chloride
- _____ d. AlCl_3
- _____ e. Boron trifluoride
- _____ f. Methyl cation
- _____ g. Methyl anion
- _____ h. Lewis base
- _____ i. Hydrogen ion
- _____ j. H^+

Modern Chemistry Chapter 6 Test Answers

Nicholas D. Tzimopoulos



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