

CHEMISTRY PAPER-II

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

OBJECTIVE

Note: You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?
 (A) *Be* (B) *Ra* (C) *Ba* (D) *Rn*
- (2) Which element forms an ion with charge +3?
 (A) *Be* (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:
 (A) *N₂O* (B) *N₂O₃* (C) *N₂O₄* (D) *N₂O₅*
- (4) Chlorine heptaoxide (*Cl₂O₇*) reacts with water to form:
 (A) Hypochlorous acid (B) Chloric acid (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO₄* is:
 (A) *ClO₃* (B) *ClO₂* (C) *Cl₂O₅* (D) *Cl₂O₇*
- (6) Which of the following is a non-typical transition element?
 (A) *Cr* (B) *Mn* (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:
 (A) Position isomerism (B) Functional group isomerism (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:
 (A) Halogenation (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?
 (A) *AlCl₃* (B) *HNO₃* (C) *BeCl₂* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?
 (A) *E₁* and *E₂* (B) *E₂* and *S_N2* (C) *S_N1* and *E₂* (D) *E₁* and *S_N1*
- (11) Which compound is more soluble in water?
 (A) *C₂H₅OH* (B) *C₆H₅OH* (C) *CH₃COCH₃* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?
 (A) Methanal (B) Ethanal (C) Propanal (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?
 (A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?
 (A) $\frac{H_2}{Ni}$ (B) $\frac{H_2}{Pt}$ (C) *NaBH₄* (D) *LiAlH₄*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:
 (A) Alcoholic *KOH* (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO₃*
- (16) Most concentrated solid nitrogen fertilizer is:
 (A) *NH₃* (B) *(NH₄)₂HPO₄* (C) *(NH₂)₂CO* (D) *NH₄NO₃*
- (17) Mark the correct statement.
 (A) *Na⁺* is smaller than *Na* atom (B) *Na⁺* is larger than *Na* atom
 (C) *Cl⁻* is smaller than *Cl* atom (D) *Cl⁻* ion and *Cl* atom are equal in size

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KJ Lindholm-Leary



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Table of Contents Chemistry Mcqs For Class 9 With Answers

1. Understanding the eBook Chemistry Mcqs For Class 9 With Answers
 - The Rise of Digital Reading Chemistry Mcqs For Class 9 With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Mcqs For Class 9 With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Platform
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Mcqs For Class 9 With Answers
 - Personalized Recommendations
 - eBook Recommendations User Reviews and Ratings
 - eBook Recommendations and Bestseller Lists

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 - Chemistry Mcqs For Class 9 With Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Chemistry Mcqs For Class 9 With Answers
 - Interactive Elements Chemistry Mcqs For Class 9 With Answers
8. Staying Engaged with Chemistry Mcqs For Class 9 With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Mcqs For Class 9 With Answers
9. Balancing eBooks and Physical Books Chemistry Mcqs For Class 9 With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Mcqs For Class 9 With Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Mcqs For Class 9 With Answers
 - Setting Reading Goals Chemistry Mcqs For Class 9 With Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Mcqs For Class 9 With Answers
 - Fact-Checking eBook Content of Chemistry Mcqs For Class 9 With Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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