

Chemistry Final Exam Study Guide 2021-22

The final exam will cover chapters 6, 7, 8 and 9, with a few very basic questions from chapters 10 and 11. Recommended ways to prepare for final exam.

- **review objectives, notes, worksheets, charts, quizzes and exams.**
- **make your own study guide**
- if you missed questions on test or quizzes, reread those topics in the textbook
- practice problems from each chapter. Appendix D in Pearson Chemistry has a problem bank with many of the answers included in the appendix. Use these for practice
- use quizzes for reviewing vocabulary, naming compounds, or other topics, most chapters of Pearson have quizzes already made or you can make your own... I have reviewed the quizzes listed on your powerschool pages and listed only those I felt were accurate.
- Study buddies, quizzing classmates or other students taking chemistry
- Take the standardized test in the textbook at the end of each chapter. I will post the answers to each of these

The exam will consist of 65-80 questions, all multiple choice. There will be copies of the periodic table, polyatomic ion chart, and transition metal charts attached to the back of the exam. You will need your calculator to complete the final exam.

Chapter 6: The Periodic Table (about 15 questions)

Vocabulary

proton
electron
atomic number
period
group
metal
nonmetal
representative element (Group A)
transition metal (Group B)
halogens
noble gas

Objectives (see p. 185 6.1 and 6.2)

- Know how the periodic table is arranged.
- Be able to use the periodic table to locate alkali metals, Group A elements, Group B elements and periods.
- Differentiate between metals, metalloids, and nonmetals.
- Be able to explain why noble gases can exist as single atoms, and usually don't react with other elements.
- **Know how to determine the number of valence electrons in a Group A element.**

Chemistry Final Exam Study Guide Multiple Choice

**Kathleen M. Savage, Charity Anne
Dorgan**



Chemistry Final Exam Study Guide Multiple Choice:

Resources in Education, 1988 **Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems Complete Book of Graduate Programs in the Arts and Sciences Princeton Review (Firm), 2004-09 Our Best 357 Colleges is the best selling college guide on the market because it is the voice of the students Now we let graduate students speak for themselves too in these brand new guides for selecting the ideal business law medical or arts and humanities graduate school It includes detailed profiles rankings based on student surveys like those made popular by our Best 357 Colleges guide as well as student quotes about classes professors the social scene and more Plus we cover the ins and outs of admissions and financial aid Each guide also includes an index of all schools with the most pertinent facts such as contact information And we've topped it all off with our school says section where participating schools can talk back by providing their own profiles It's a whole new way to find the perfect match in a graduate school **Class 11-12 Chemistry MCQ (Multiple Choice Questions)** Arshad Iqbal, 2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ

Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental particles Heisenberg s uncertainty principle hydrogen spectrum magnetic quantum number mass of electron metallic crystals properties Moseley law neutron properties orbital concept photons wave number Planck s quantum theory properties of cathode rays properties of positive rays quantum numbers quantum theory Rutherford model of atom shapes of orbitals spin quantum number what is spectrum x rays and atomic number The Basic Chemistry MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Basic chemistry atomic mass atoms molecules Avogadro s law combustion analysis empirical formula isotopes mass spectrometer molar volume molecular ions moles positive and negative ions relative abundance spectrometer and stoichiometry The Chemical Bonding MCQ PDF e Book Chapter 3 practice test to solve MCQ questions on Chemical bonding chemical combinations atomic radii atomic radius periodic table atomic ionic and covalent radii atoms and molecules bond formation covalent radius electron affinity electronegativity electronegativity periodic table higher ionization energies ionic radius ionization energies ionization energy periodic table Lewis concept and modern periodic table The Experimental Techniques MCQ PDF e Book Chapter 4 practice test to solve MCQ questions on Experimental techniques chromatography crystallization filter paper filtration filtration crucibles solvent extraction and sublimation The Gases MCQ PDF e Book Chapter 5 practice test to solve MCQ questions on Gas laws gas properties kinetic molecular theory of gases ideal gas constant ideal gas density liquefaction of gases absolute zero derivation applications of Daltons law Avogadro s law Boyle s law Charles law Daltons law diffusion and effusion Graham s law of diffusion ideality deviations kinetic interpretation of temperature liquids properties non ideal behavior of gases partial pressure calculations plasma state pressure units solid s properties states of matter thermometry scales and van der Waals equation The Liquids and Solids MCQ PDF e Book Chapter 6 practice test to solve MCQ questions on Liquid crystals types of solids classification of solids comparison in solids covalent solids properties of crystalline solids Avogadro number determination boiling point external pressure boiling points crystal lattice crystals and classification cubic close packing diamond structure dipole dipole forces dipole induced dipole forces dynamic equilibrium energy changes intermolecular attractions hexagonal close packing hydrogen bonding intermolecular forces London dispersion forces metallic crystals properties metallic solids metal s structure molecular solids phase changes energies properties of covalent crystals solid iodine structure unit cell and vapor pressure Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 Study Guide for Introductory Chemistry Iris Stovall,1990 **Clinical Medicine; Examination Review** George Horemis,1972 **Guide to Science**

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