

Chapter Test B

Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The gas volume of a diatomic molecule with a molar mass of 28 g/mol is 1.12 L at STP. How many molecules of the gas are there?
 - gas
 - grams
 - moles
 - liters
- The volume of a gas at STP is 1.12 L. How many molecules of the gas are there?
 - moles
 - grams
 - liters
 - atoms
- The volume of a gas at STP is 1.12 L. How many molecules of the gas are there?
 - atoms
 - grams
 - moles
 - liters
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 - atoms
 - grams
 - moles
 - liters

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**Darrell D. Ebbing, Larry K.
Krannich, Joan I. Senyk**



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