

Solubility Product Constant (K_{sp})



$$K_{sp} = [M^{y+}]^x [N^{x-}]^y$$

$[M^{y+}] \rightarrow$ Concentration of M^{y+}

$[N^{x-}] \rightarrow$ Concentration of N^{x-}

Solubility Product Constant Lab Activity

M Mosston



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