

f-block elements \div Group - 8, Period $\begin{cases} 6 - \text{lanthanoids} \\ 7 - \text{actinoids} \end{cases}$

\rightarrow are those in

which the 4f and 5f orbitals are progressively filled

\rightarrow Placed below the table in two long periods (6th, 7th)

\rightarrow Formal members of group 8, but taken out

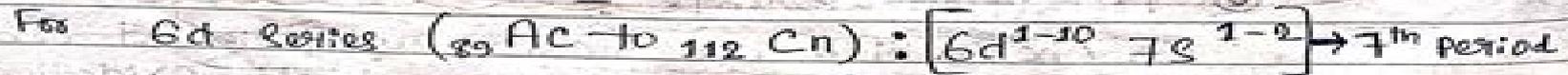
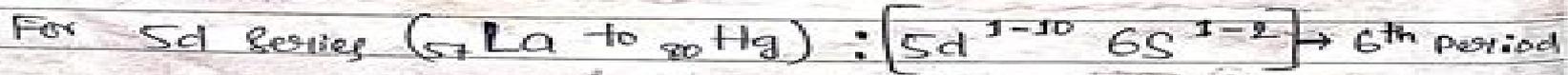
\rightarrow Inner transition metals

Lanthanoids and Actinoids.

The Transition Elements \div (d-block)

The very name transition given to the elements of d-block is only because of their position between s and p block elements.

\rightarrow General E.C is



Inorganic Chemistry Notes For Iit Jee

Raju Sharma



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