

CALCULATION TITLE

	6	Uniformly Normalized	
	mole %	mole %	
C0H4	93.42	93.7417	%
C2H6	0.05	0.0512	%
C3H8	0	0.0000	%
C4H10	0	0.0000	%
C4H12	0	0.0000	%
C5H12	0.02	0.0164	%
C6H14	0.01	0.0072	%
C7H16	0	0.0000	%
C8H18	0	0.0000	%
C9H20	0	0.0000	%
C10H22	0	0.0000	%
C6H6	0	0.0000	%
CO	0	0.0000	%
CO2	2.29	2.3490	%
H2S	0	0.0000	%
H2=O2	0	0.0000	%
H2	0	0.0000	%
O2	0	0.0000	%
N2	1.79	1.8345	%
H2O	0	0.0000	%
	97.5750	100.0000	%

To carry out calculations for US pressure and temperature units, select the "Calculate" button to link from the US to the SI units sheet. Enter details and data into the cells with a yellow background only. Real Cp, Cv, k and WGS are calculated using the Redlich-Kwong EOS. Select the "Reset" button to relink from the SI to the US units sheet.

SI Calc Active

R=	1.0859	kJ/kg.mol.degK		
MW av	28.96	lb/lb.mol		
Zb av	1			
Pb	0	psia(g)		0.0 kPa(g)
Pb	14.7	psia(a)		
Pb	14.696	psia(a)		
Tb	59	degF		15.0 degC
Tb	459.67	degR		
Tb	518.67	degR		

MW	18.9409	lb/lb.mol		
SG ideal	0.5849	Specific Gravity		
SG real	0.5859			
Zb (20°F, 0 psia(g))	0.9499	Compressibility at base conditions		
HHV ideal	971.262	Btu/Sft3	Higher Heating Value	
HHV real	973.141	Btu/Sft3	Wobbe Index	47.35
LHV ideal	874.424	Btu/Sft3	Lower Heating Value	
LHV real	876.170	Btu/Sft3		
P	14.696	psia(g)	Operating Pressure	10000.0 psia(g)
T	59	°F	Operating Temperature	30.0 °C
P	1469.1	psia(a)		
T	545.67	°R		
Pc	665.96	psia(a)	Pseudo Critical Pressure	includes Wichert & Aziz Correction for Sour Gas
Tc	342.11	°R	Pseudo Critical Temperature	includes Wichert & Aziz Correction for Sour Gas
Vc	0.0934	ft3/lb	Pseudo Critical Volume	
Zc	0.2879		Pseudo Critical Compressibility	
Pr (Pre)	2.2003	0.0221	Pseudo Reduced Pressure	
Tr	1.5959		Pseudo Reduced Temperature	
Z (MW=40, P=10150 psia)	#VALUE!		Compressibility (Pr 0 to 15, Tr 1 to 3)	#VALUE! from AGAS GCM2 Method
ρ	#VALUE!	lb/ft3	Density	#VALUE! from AGAS GCM2 Method
Cpo (T =25 to 150°C)	#VALUE!	Btu/lb-mole-°F	Heat Capacity (P=constant)	#VALUE! Btu/lb-°F
Cvo	#VALUE!	Btu/lb-mole-°F	Heat Capacity (T=constant)	#VALUE! Btu/lb-°F
k=Cpo/Cvo	#VALUE!		Thermal Conductivity	#VALUE! Btu/hr-ft-°F
Cp	#VALUE!	Btu/lb-mole-°F	refer Redlich-Kwong	#VALUE! Btu/lb-°F
Cv	#VALUE!	Btu/lb-mole-°F	Equations of State	#VALUE! Btu/lb-°F
k=Cp/Cv	#VALUE!		Velocity of Sound	#VALUE! ft/sec
ω	0.0174		Acentric Factor	
μ (T =202 to 932°F)	#VALUE!	cP	Viscosity from Fig 23-23 (Pr 0 to 3040 psia)	#VALUE! from John M Campbell & Co
μa @14.696 psia(a)	#VALUE!	cP	or Viscosity from 23-22 (if Tr > 1.0)	
μva	#VALUE!	cP	and pressure correction from 23-24	
μ	#VALUE!	cP		
or if high levels of non hydrocarbons are present (eg 20%)				
μ	0.04555		Eq 23-20 (corrected)	
μa	0.00051		Eq 23-21 (corrected) and Eq 23-22	
μa @14.696 psia(a)	0.01121	cP	for high levels of non hydrocarbons	
μva	#VALUE!	cP	with pressure correction from 23-24	
H (T =22 to 895°F)	#VALUE!	Btu/lb-mole	Enthalpy (Pr 0.1 to 15, Tr 1 to 4)	Note:H=0 for T=0°R and P=0 psia(a)
UT	#VALUE!	Btu/lb	Note: Values for Hexane, Heptane, Octane, Nonane, Decane are approximate only!	
JT	#VALUE!	°F/100 psi	JT Effect of Gas Expansion	
S (T =112 to 464°F)	#VALUE!	Btu/lb-mole-°F	Entropy (Pr 0.1 to 10, Tr 0.8 to 4)	Note:S=0 for T=0°R and P=0 psia(a)
	#VALUE!	Btu/lb-°F	Note: Values for Hexane, Heptane, Octane, Nonane, Decane are not available!	

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