

CALCULATION TITLE

	6	Uniformly Normalized
	mole %	mole %
C044	Methane	93.42
C246	Ethane	0.05
C348	Propane	0
C4410	i-Butane	0
C4410	n-Butane	0
C5412	i-Pentane	0
C5412	n-Pentane	0.02
C6414	n-Hexane	0.01
C7416	n-Heptane	0
C8418	n-Octane	0
C9420	n-Nonane	0
C10422	n-Decane	0
C646	Benzene	0
CO	Carbon Monoxide	0
CO2	Carbon Dioxide	2.29
H2S	Hydrogen Sulfide	0
H2=O2	Air	0
H2	Hydrogen	0
O2	Oxygen	0
N2	Nitrogen	1.79
H2O	Water	0
	67.5750	100.0000

To carry out calculations for US pressure and temperature units, select the "Calculate" button to link from the US to the SI units sheet. Enter details and data into the cells with a yellow background only. Real Cp, Cv, k and WGS are calculated using the Redlich-Kwong EOS. Select the "Reset" button to relink from the SI to the US units sheet.

SI Calc Active

R=	1.0859	kJ/kg.mol.degK		
MW av	28.96	lb/lb.mol		
Zo av	1			
Po	0	psia(g)		0.0 kPa(g)
Pr	14.7	psia(a)		
Po	14.696	psia(a)		
To	59	degF		15.0 degC
Tr	459.67	degR		
To	518.67	degR		

MW	18.9409	lb/lb.mol	
SG ideal	0.5849	Specific Gravity	
SG real	0.5859		
Zo (20°F, 0 psia(g))	0.9499	Compressibility at base conditions	
HHV ideal	971.262	Btu/Sft3	Higher Heating Value
HHV real	973.141	Btu/Sft3	Wobbe Index
LLHV ideal	874.424	Btu/Sft3	Lower Heating Value
LLHV real	876.170	Btu/Sft3	
P	14.696	psia(g)	Operating Pressure
T	59	°F	Operating Temperature
P	1469.1	psia(a)	
T	545.67	°R	
Pc	665.98	psia(a)	Pseudo Critical Pressure
Tc	342.11	°R	Pseudo Critical Temperature
Vc	0.0934	ft3/lb	Pseudo Critical Specific Volume
Zc	0.2879		Pseudo Critical Compressibility
Pr (Proc)	2.2003	0.0221	Pseudo Reduced Pressure
Tr	1.5959		Pseudo Reduced Temperature
Z (MW=40, P=10150 psi)	#VALUE!		Compressibility (Pr 0 to 15, Tr 1 to 3)
ρ	#VALUE!	lb/ft3	Density
Cpo (T =25 to 150°C)	#VALUE!	Btu/lb-mole-°F	Heat Capacity (P constant)
Cvo	#VALUE!	Btu/lb-mole-°F	Heat Capacity (T constant)
k=Cpo/Cvo	#VALUE!		Thermal Conductivity
Cp	#VALUE!	Btu/lb-mole-°F	refer Redlich-Kwong
Cv	#VALUE!	Btu/lb-mole-°F	Equations of State
k=Cp/Cv	#VALUE!		Velocity of Sound
ω	0.0174		Acentric Factor
μ (T =202 to 932°F)	#VALUE!	cP	Viscosity from Fig 23-23 (Pr 0 to 3040 psi)
μa @14.696 psia(a)	#VALUE!	cP	or Viscosity from 23-22 (if Tr > 1.0)
μva	#VALUE!	cP	and pressure correction from 23-24
μ	#VALUE!	cP	
or if high levels of non hydrocarbons are present (eg 20%)			
μ	0.04555		Eq 23-20 (corrected)
μa	0.00251		Eq 23-21 (corrected) and Eq 23-22
μa @14.696 psia(a)	0.01121	cP	for high levels of non hydrocarbons
μva	#VALUE!	cP	with pressure correction from 23-24
H (T =22 to 895°F)	#VALUE!	Btu/lb-mole	Enthalpy (Pr 0.1 to 15, Tr 1 to 4)
HT	#VALUE!	Btu/lb	Note: Values for Hexane, Heptane, Octane, Nonane, Decane are approximate only!
JT	#VALUE!	°F/100 psi	JT Effect of Gas Expansion
S (T =112 to 464°F)	#VALUE!	Btu/lb-mole-°F	Entropy (Pr 0.1 to 10, Tr 0.8 to 4)
S	#VALUE!	Btu/lb-°F	Note: Values for Hexane, Heptane, Octane, Nonane, Decane are not available!

#VALUE! from AGAS GCM2 Method

10000.0 psia(g)

20.0 °C

includes Wichert & Aziz Correction for Sour Gas

includes Wichert & Aziz Correction for Sour Gas

#VALUE! from AGAS GCM2 Method

#VALUE! from AGAS GCM2 Method

#VALUE! Btu/lb-°F

#VALUE! from John M Campbell & Co

Note: H=0 for T=0°R and P=0 psia(a)

Note: Values for Hexane, Heptane, Octane, Nonane, Decane are approximate only!

Note: S=0 for T=0°R and P=0 psia(a)

Note: Values for Hexane, Heptane, Octane, Nonane, Decane are not available!

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