

Please answer each question carefully. Remember, follow your gut! Please copy all answers on the answer sheet on the back of this exam! Otherwise, I will take off 10 points!

- The structure of DNA described by Watson and Crick included:
  - a double helix
  - the sugar-phosphate backbone aligned in the center of the helix
  - the base pairs that are stacked on the inside of the double helix
  - A double helix and base pairs that are stacked on the inside of the double helix
- What gives proteins such a dominant role in biochemistry?
  - the ability to act as a blueprint
  - their ability to spontaneously fold into complex three-dimensional structures
  - the variation in protein sizes
  - their ability to self-replicate
- Match the loss of a particular organelle with the associated disease.
  - Hypercholesterolemia - smooth endoplasmic reticulum
  - Diabetes - endosome
  - Tay-Sachs disease - lysosome
  - Muscle degeneration - mitochondria
- In a biochemistry lab course, you are asked to design an experiment to identify a strain of bacteria. Your lab partner claims that she thinks the bacterium contains a rough endoplasmic reticulum. To verify her claim, which of the following experiments would you perform? (pretend you're indulging her ludicrous claim).
  - determine whether the bacterium has an internal membrane-enclosed compartment
  - determine whether the bacterium generates  $\text{CO}_2$  in the presence of fuel molecules
  - determine whether the bacterium can synthesize proteins
  - determine whether the bacterium can synthesize ATP in the presence of fuel molecules and  $\text{O}_2$
- What is the  $\text{H}^+$  concentration in a urine sample that has a pH of 6?
  - $10^{-6} \text{ M}$
  - $10^{-8} \text{ M}$
  - $10^6 \text{ M}$
  - $10^{-14} \text{ M}$
- Which of the following is considered a noncovalent bond?
  - electrostatic interactions
  - hydrogen bonds
  - van der Waals interactions
  - All of these answers
- Which functional group is shown below?



- Amine
  - Thiol
  - Carboxylic acid
  - None of these answers
- What is the  $[\text{A}^-]/[\text{HA}]$  ratio when the weak acid is in a solution one pH unit above its  $\text{pK}_a$ ?

# Biochemistry Quiz Questions And Answers

**L Towne**



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Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current

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**Oxford Dictionary of English** Angus Stevenson,2010-08-19 19 pages of contents in middle of book between end of L and beginning of M

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