

CBSE Class 12 Chemistry Sample Paper Set 8

CHEMISTRY
CLASS-XII

Time Allotted: 3 Hrs

Maximum Marks: 70

General Instructions:

1. All questions are compulsory.
2. Question No. 1-8 are very short answer questions and carry 1 mark each.
3. Question No. 9-18 are short answer questions and carry 2 marks each.
4. Question No. 19-27 are also short answer questions and carry 3 marks each.
5. Question No. 28-30 are long answer questions and carry 5 marks each.
6. Use log tables if necessary, use of calculators is not allowed.

Q.1 Bond enthalpy of fluorine is lower than that of chlorine, why?

Q.2 Give the IUPAC name of the following compound:



Q.3 On increasing temperature, activation energy of a reaction decreases, why?

Q.4 Which of the following is most effective electrolyte in the coagulative of AgI/Ag+sol?
 K_2SO_4 , MgCl_2 , $\text{K}_4[\text{Fe}(\text{CN})_6]$

Q.5 Write the reaction when glucose is heated with excess of HI.

Q.6 Which Xe compound has distorted octahedral shape?

Q.7 What is the denticity of co-ordination compound used for the treatment of lead Poisoning?

Q.8 An alkoxide is a stronger base than hydroxide ion. Justify.

Q.9 (a) State the law which helps to determine the limiting molar conductivity of weak electrolyte.

(b) Calculate limiting molar conductivity of CaSO_4 (limiting molar conductivity of calcium and sulphate ions are 119.0 and 160.0 $\text{S cm}^2 \text{mol}^{-1}$ respectively).

Q.10 Rate constant K for first order reaction has been found to be $2.54 \times 10^{-3} \text{ sec}^{-1}$. Calculate its three-fourth life.

OR

A first order gas reaction $\text{A}_2(\text{g})\text{B}_2(\text{g}) \rightarrow 2\text{A}(\text{g}) + 2\text{B}(\text{g})$ at the temperature 400°C has the rate Constant $K = 2.0 \times 10^{-4} \text{ sec}^{-1}$. What percentage of A_2B_2 is decomposed on heating for 900 seconds.

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Table of Contents 12th Chemistry Practice Paper Doc

1. Understanding the eBook 12th Chemistry Practice Paper Doc
 - The Rise of Digital Reading 12th Chemistry Practice Paper Doc
 - Advantages of eBooks Over Traditional Books
2. Identifying 12th Chemistry Practice Paper Doc
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12th Chemistry Practice Paper Doc
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Chemistry Practice Paper Doc
 - Personalized Recommendations
 - 12th Chemistry Practice Paper Doc User Reviews and Ratings
 - 12th Chemistry Practice Paper Doc and Bestseller Lists
5. Accessing 12th Chemistry Practice Paper Doc Free and Paid eBooks

- 12th Chemistry Practice Paper Doc Public Domain eBooks
 - 12th Chemistry Practice Paper Doc eBook Subscription Services
 - 12th Chemistry Practice Paper Doc Budget-Friendly Options
6. Navigating 12th Chemistry Practice Paper Doc eBook Formats
 - ePub, PDF, MOBI, and More
 - 12th Chemistry Practice Paper Doc Compatibility with Devices
 - 12th Chemistry Practice Paper Doc Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Chemistry Practice Paper Doc
 - Highlighting and Note-Taking 12th Chemistry Practice Paper Doc
 - Interactive Elements 12th Chemistry Practice Paper Doc
 8. Staying Engaged with 12th Chemistry Practice Paper Doc
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Chemistry Practice Paper Doc
 9. Balancing eBooks and Physical Books 12th Chemistry Practice Paper Doc
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Chemistry Practice Paper Doc
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine 12th Chemistry Practice Paper Doc
 - Setting Reading Goals 12th Chemistry Practice Paper Doc
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of 12th Chemistry Practice Paper Doc
 - Fact-Checking eBook Content of 12th Chemistry Practice Paper Doc
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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