

Single Replacement Reaction Stoichiometry

Hina K. Huynh

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Purpose:

To learn how to balance chemical reactions as well as determining the limiting reactant and theoretical yield. Also, be able to analyze the product of chemical reactions.

Procedure:

During this experiment, an amount of aluminum foil will be mixed with CuSO_4 or copper sulfate to see how the two chemicals react. The reaction will be observed in the production of copper metal.

Data

Activity 1

Data Table 1

Write the balanced Chemical Equation:



	<i>Answer (include the units)</i>	<i>Observations</i>	<i>Calculations</i>
Volume of 1 M CuSO_4	<i>100.0 mL</i>	<i>measured on meniscus line</i>	<i>measured with a graduated cylinder</i>

Solution Stoichiometry Lab Report

Valentine Korah



Solution Stoichiometry Lab Report:

Solution Stoichiometry Lab Report Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Solution Stoichiometry Lab Report**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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