

# DIGITAL HAND TALLY COUNTER



# Manual Numbers Counting

**G Thomas**



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element therefore differ from each other only in the number of neutrons within the nucleus

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protons isotopes are atoms that have the same number of protons but different numbers of neutrons thus atoms of 6 12 c and  
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neutrons differ ions are atoms which contain an overall charge where number of protons number of electrons

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example 99 of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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