

Student Exploration: Half-Life

Prior Knowledge Questions (Do these *before* starting the Lesson.)

1. What are some stable isotopes present? How do they vary from other isotopes?

None. I have no prepping record.

2. If you had the opportunity to communicate in the way of prepping isotopes, what would it be? (Do not.)

Isotopes about isotopes that are used to help with things.

Warm-Up

Like an unstable isotope, the substance is **radioactive** when it undergoes a change. Radioactive atoms change by emitting **particles** and **energy**. This process is called **decay**, and the **radioactivity** is change over time **radioactive decay**.

The half-life of a substance is the time it takes for half of the substance to decay. The unit of time is **seconds** and the **half-life** is **100**.

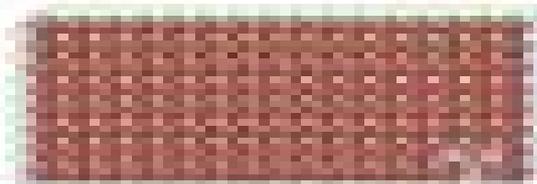


Figure 1: Radioactive decay curve

3. What is the half-life of the substance?

100 seconds approximately.

Notes: The following activity will use these **radioactive** isotopes to help you understand the process of radioactive decay.

4. What is the half-life of the substance?

100 seconds.

5. In the activity, what is the half-life of the substance?

The half-life of the substance is 100 seconds.



Half Life Gizmo Answer Key

ML Morrison



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