

# CHAPTER 3 CHAPTER ASSESSMENT

## MATTER AND CHANGE

### Reviewing Vocabulary

Match the definition in Column A with the term in Column B.

#### Column A

- \_\_\_\_\_ 1. The outermost electrons of an atom
- \_\_\_\_\_ 2. The center of an atom
- \_\_\_\_\_ 3. Hot, highly ionized, electrically conducting gas
- \_\_\_\_\_ 4. The smallest particle of an element that retains that element's characteristics
- \_\_\_\_\_ 5. The change from a gas to a liquid
- \_\_\_\_\_ 6. A homogeneous mixture
- \_\_\_\_\_ 7. The change of one or more substances into other substances
- \_\_\_\_\_ 8. The attractive force between two ions of opposite charge
- \_\_\_\_\_ 9. A tiny particle with a negative electrical charge
- \_\_\_\_\_ 10. The average of the mass numbers of an element's isotopes
- \_\_\_\_\_ 11. The spontaneous process through which unstable nuclei emit radiation

#### Column B

- a. atom
- b. atomic mass
- c. chemical reaction
- d. condensation
- e. electron
- f. ionic bond
- g. solution
- h. nucleus
- i. plasma
- j. valence electrons
- k. radioactive decay

In the space at the left, write the word or phrase in parentheses that makes the statement correct.

- \_\_\_\_\_ 12. The combined number of protons and neutrons is the (atomic number, mass number).
- \_\_\_\_\_ 13. An (energy level, isotope) represents the area in an atom where an electron is most likely to be found.
- \_\_\_\_\_ 14. (An ionic bond, A covalent bond) is the attraction of two atoms for a shared pair of electrons that holds the atoms together.
- \_\_\_\_\_ 15. An atom that gains or loses an electron is a charged particle called (an ion, a molecule).
- \_\_\_\_\_ 16. (An element, A compound) is a substance that cannot be broken down into simpler substances.
- \_\_\_\_\_ 17. A tiny particle that has mass and a positive electrical charge is a (neutron, proton).
- \_\_\_\_\_ 18. (Acids, Bases) are solutions characterized by the formation of hydroxide ions ( $\text{OH}^-$ ).
- \_\_\_\_\_ 19. The process of changing from a liquid into a gas is called (evaporation, sublimation).

# Chemistry Matter And Change Chapter 7

**Jicheng Xie**



## **Chemistry Matter And Change Chapter 7:**

## Reviewing **Chemistry Matter And Change Chapter 7**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemistry Matter And Change Chapter 7**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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