

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct answer in the blank.

Column I

- _____ 1. The volume of a fixed amount of gas varies inversely as the pressure of the gas, provided the temperature is constant.
- _____ 2. The volume of a fixed amount of gas varies directly as the absolute temperature of the gas, provided the pressure is constant.
- _____ 3. The volume of a fixed amount of gas varies directly as the absolute temperature of the gas, provided the pressure is constant.
- _____ 4. The volume of a fixed amount of gas varies directly as the absolute temperature of the gas, provided the pressure is constant.
- _____ 5. A gas exerts pressure on the walls of its container because of the collisions of the gas molecules with the walls.
- _____ 6. The average kinetic energy of gas molecules is proportional to the absolute temperature of the gas.
- _____ 7. The average kinetic energy of gas molecules is proportional to the absolute temperature of the gas.
- _____ 8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
- _____ 9. The average kinetic energy of gas molecules is proportional to the absolute temperature of the gas.
- _____ 10. The average kinetic energy of gas molecules is proportional to the absolute temperature of the gas.

Column II

- a. Boyle's law
- b. Charles' law
- c. Avogadro's law
- d. Gay-Lussac's law
- e. Boyle's law
- f. Boyle's law
- g. Charles' law
- h. Boyle's law
- i. Charles' law
- j. Boyle's law
- k. Boyle's law
- l. Boyle's law
- m. Boyle's law
- n. Boyle's law
- o. Boyle's law
- p. Boyle's law
- q. Boyle's law
- r. Boyle's law
- s. Boyle's law
- t. Boyle's law
- u. Boyle's law
- v. Boyle's law
- w. Boyle's law
- x. Boyle's law
- y. Boyle's law
- z. Boyle's law

B. Multiple Choice

Circle the correct answer and write its letter in the blank.

- _____ 1. According to Boyle's law, if the volume of a fixed amount of gas is doubled, the pressure of the gas will be:
 - a. halved
 - b. doubled
 - c. quadrupled
 - d. halved

Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers

Karin Nielsen-Saines



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