

3. Which oxides cause acid rain ? What is its pH value ?
4. The empirical formula of a compound is CH_2O . Its molecular weight is 90. Calculate the molecular formula of the compound.
5. What is Chemical Oxygen Demand (COD) ?
6. How does graphite function as a lubricant ?
7. How is water gas prepared ?
8. State the first law of the thermodynamics.
9. What is the relationship between C_p and C_v ?
10. Write IUPAC names of the following compounds :



SECTION B

6×4=24

Note :—Answer ANY SIX questions.

11. Deduce :
 - (a) Boyle's law, and
 - (b) Charles's law
 from kinetic gas equation.
12. Explain the hybridization involved in PCl_5 molecule.
13. Explain the formation of coordinate covalent bond with one example.
14. Balance the following redox reaction by Ion-electron method :

$$\begin{array}{ccccccc} \text{MnO}_4^- & + & \text{SO}_2 & \rightarrow & \text{Mn}^{2+} & + & \text{HSO}_4^- \\ \text{(aq)} & & \text{(g)} & & \text{(aq)} & & \text{(aq)} \end{array}$$

(acidic solution)
15. Discuss the application of Le-Chatelier's principle for the synthesis of Ammonia by Haber's process.
16. What is hard water ? Write a note on the Calgon's method for the removal of hardness of water.

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