

## Chapter Test II

### A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

#### Column I

1. An unbalanced chemical equation for a reaction that obeys the law of conservation of matter.
2. The number of protons in an atom of an element. It is equal to the number of electrons.
3. The number of electrons in an atom of an element.
4. The ratio of the actual weight of an element to the standard weight of 16 grams.
5. The number of protons and neutrons in an atom of an element. It is equal to the atomic number.
6. The number of protons and neutrons in an atom of an element. It is equal to the atomic number.

#### Column II

- a. atomic weight
- b. limiting reagent
- c. molecular weight
- d. stoichiometry
- e. atomic number
- f. atomic weight

### B. Multiple Choice

Choose the correct answer for each question.

1. The number of protons in an atom of an element is equal to the number of electrons. This is true for all elements.
  - a. true for all elements
  - b. true for all elements
  - c. true for all elements
  - d. true for all elements
2. The number of protons in an atom of an element is equal to the number of electrons. This is true for all elements.
  - a. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$ , 5 protons in  $\text{B}$
  - b. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$
  - c. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$
  - d. none of these
3. The number of protons in an atom of an element is equal to the number of electrons. This is true for all elements.
  - a. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$
  - b. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$
  - c. 1 proton in  $\text{H}$ , 2 protons in  $\text{He}$ , 3 protons in  $\text{Li}$ , 4 protons in  $\text{Be}$
  - d. none of these

# Chemistry Chapter 12 Stoichiometry Test

**L Darling-Hammond**



**Chemistry Chapter 12 Stoichiometry Test:**

## **Chemistry Chapter 12 Stoichiometry Test** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Chemistry Chapter 12 Stoichiometry Test**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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