

Chapter 8

Solutions, Acids, and Bases

Section 8.1 Formation of Solutions

(pages 128-133)

Three factors determine the extent of a solution: the processes that occur when components interact, whether the properties of a mixture compare with those of its components, and solubility.

Properties of Liquid Solutions (page 128)

1. **Thermodynamic properties of a solution can differ from those of its components.**
 - a. _____, _____, _____
 - b. _____, _____, _____
 - c. _____, _____, _____
2. **Comparative enthalpies of two substances in a gas solution:**
_____ and _____ differ due to a poor description of gaseous mixture and concept of a good description of chemical formula.
3. **Order into the water molecules that hydrogen bonds are formed:**
 - a. The water molecules become more organized.
 - b. The water molecules become more disorganized.
 - c. The water molecules relax.
 - d. The water molecules are not affected by hydrogen bonds.

Heat of Solution (page 128)

1. **Calculating enthalpy changes for solution steps:** _____, _____
processes are exothermic heat.
2. **Calculating enthalpy change for a solution:** _____, _____
processes are exothermic heat.
3. **For the following processes that are exothermic, the enthalpy change is negative and the substances strongly attract particles. How is this related to the process?**

Chapter 8 Solutions Acids Bases Section 82 Answers

K. Peter C. Vollhardt, Neil E. Schore



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