

NUCLEAR CHEMISTRY

BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction (α emission, β emission, γ emission, positron emission, artificial transmutation, fission, or fusion) described.

	Type of Nuclear Reaction
1.) $^{42}_{19}\text{K} \rightarrow ^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	1.)
2.) $^{239}_{94}\text{Pu} \rightarrow ^{4}_{2}\text{He} + \underline{\hspace{2cm}}$	2.)
3.) $^{235}_{92}\text{U} \rightarrow \underline{\hspace{2cm}} + ^{231}_{90}\text{Th}$	3.)
4.) $^1_1\text{H} + ^2_1\text{H} \rightarrow \underline{\hspace{2cm}}$	4.)
5.) $^6_3\text{Li} + ^1_0\text{n} \rightarrow ^4_2\text{He} + \underline{\hspace{2cm}}$	5.)
6.) $^{27}_{13}\text{Al} + ^4_2\text{He} \rightarrow ^{30}_{15}\text{P} + \underline{\hspace{2cm}}$	6.)
7.) $^9_4\text{Be} + ^1_1\text{H} \rightarrow \underline{\hspace{2cm}} + ^4_2\text{He}$	7.)
8.) $^{37}_{19}\text{K} \rightarrow ^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	8.)
9.) $\underline{\hspace{2cm}} + ^1_0\text{n} \rightarrow ^{142}_{56}\text{Ba} + ^{91}_{36}\text{Kr} + 3\ ^1_0\text{n}$	9.)
10.) $^{238}_{92}\text{U} + ^4_2\text{He} \rightarrow \underline{\hspace{2cm}} + ^1_0\text{n}$	10.)
11.) $^{14}_6\text{C} \rightarrow ^{14}_7\text{N} + \underline{\hspace{2cm}}$	11.)
12.) $^{187}_{75}\text{Re} + \underline{\hspace{2cm}} \rightarrow ^{188}_{75}\text{Re} + ^1_1\text{H}$	12.)
13.) $^{22}_{11}\text{Na} + \underline{\hspace{2cm}} \rightarrow ^{22}_{10}\text{Ne}$	13.) ---
14.) $^{218}_{84}\text{Po} \rightarrow \underline{\hspace{2cm}} + ^4_2\text{He}$	14.)
15.) $^{253}_{99}\text{Es} + ^4_2\text{He} \rightarrow ^1_0\text{n} + \underline{\hspace{2cm}}$	15.)

Write the Balanced nuclear equations for the alpha decay of:

a) Plutonium-234

b) Strontium-90

Write the balanced nuclear equations for the alpha, beta and gamma decay of Radium-226

Chapter 28 Nuclear Chemistry Worksheet Answers

LL Leslie



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Resources in Education*, 1989 **Whitaker's Five-year**

Cumulative Book List, 1968 *General Catalogue of Printed Books* British Museum. Department of Printed Books, 1969

Current Index to Journals in Education, 1989

Subject Guide to Books in Print, 1975

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Through Problems Hari Jeevan Arnikar, 1997 In Most Of Our Universities The Course In Advanced Chemistry Is Open To Students Of Two Streams One Who Had Mathematics Physics And Chemistry The Mpc Group And The Other With Life Or Earth Science And Chemistry At The B Sc Stage A Problem Arises With The Students Of The Latter Stream Who Had No Background In Mathematics Beyond The High School Stage However They Cannot Be Denied Admission To Higher Chemistry Courses On This Ground All The Same These Non Mathematics Students Start Realizing Soon That They Are Missing Some Of The Essentials Of The Subject Available To The Other Fellow Students Those Of The Mpc Group Chemistry Is A Physical Science Involving Measurements Of Precision In Respect Of The Amounts Of Chemicals Reacting And Of The Amounts Of The Products Formed How Fast And How Far A Given Reaction Goes The Energy Changes Involved And The Quantitative Effects Due To Variations In The Relevant Parameters All These Interrelated Quantities Are Governed By Precise Laws Expressed In The Form Of Mathematical Equations One Cannot Be A True Master Of Chemistry In Any Branch Unless He Is Comfortably At Home With The Equations Relevant To That Branch And Can Use Them Correctly For Solving Problems Nuclear Chemistry Through Problems Is Written With The Object Of Helping The Student In Solving Numerical Problems In The Subject It Is Meant To Be A Companion To The Main Textbook Essentials Of Nuclear Chemistry Iv Ed 19 95 It Cannot Be Considered As A Substitute To The Latter The Background Material Given At The Beginning Of Each Chapter Is Necessary And Sufficient For Solving Numerical Problems After Some Practice It Is Hoped That The Student Will Be Able To Solve The Problems By Himself Without Looking Into The Solution Provided By Us Except For Checking The Final Answer Printed In Bold Type At The End Of The Solution **Nuclear Chemistry** R. K. Dave, 2009-01-01 **NUCLEAR**

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