

**STUDENT BOOK ANSWERS**
**To come**
**Chapter 9: Gas laws**
**Worked example 9.1**

- 1  $n = \frac{V}{V_m} = \frac{2.33}{22.71} = 0.1026 \text{ mol}$   
 $m = n \times MM = 0.1026 \times (12.01 + (4 \times 1.008)) = 1.646 = 1.65 \text{ g (3sf)}$
- 2  $n = \frac{m}{MM} = \frac{23.0}{14.01 + (3 \times 1.008)} = 1.350 \text{ mol}$   
 $V = n \times V_m = 1.350 \times 24.79 = 33.47 = 33.5 \text{ L (3sf)}$

**Worked example 9.2**

- 1  $2\text{Na(s)} + 2\text{H}_2\text{O(l)} \rightarrow 2\text{NaOH(aq)} + \text{H}_2\text{(g)}$   
 $n(\text{Na}) = \frac{m}{MM} = \frac{1.87}{22.99 + 16.00 + 1.008} = 0.04675 \text{ mol}$   
 $n(\text{H}_2) = \frac{1}{2} \times n(\text{Na}) = 0.5 \times 0.04675 = 0.02338 \text{ mol}$   
 $V = n \times V_m = 0.02338 \times 24.79 = 0.5795 = 0.580 \text{ L (3sf)}$
- 2  $n(\text{HCl}) = \frac{V}{V_m} = \frac{13.4}{22.71} = 0.5900 \text{ mol}$   
 $2\text{KCl(aq)} + \text{H}_2\text{SO}_4\text{(aq)} \rightarrow 2\text{HCl(g)} + \text{K}_2\text{SO}_4\text{(aq)}$   
 $n(\text{KCl}) = n(\text{HCl}) = 0.5900 \text{ mol}$   
 $m(\text{KCl}) = n \times MM = 0.5900 \times (39.10 + 35.45) = 43.98 = 44.0 \text{ g (3sf)}$

**Check your understanding 9.1/9.2**

- 1 The volume of a gas changes with temperature and pressure, both of which compress or expand the gas particles.
- 2 Use the formula  $n = \frac{V}{V_m}$  to convert the volume of gas to a number of moles.  $V_m$  depends upon the temperature of the gas. Use  $m = n \times MM$  to convert the number of moles ( $n$ ) to a mass ( $m$ ) by multiplying by the molecular mass ( $MM$ ).
- 3 **a** **i**  $2 \times 250 = 500 \text{ mL}$   
**ii**  $250 \text{ mL}$   
**b** **i**  $500 \text{ mL}$   
**ii** Nothing, as the law of combining volumes only applies to gases.

# Chapter 9 Notes Cengage Learning

**City of London College of Economics**



## Chapter 9 Notes Cengage Learning:

Governance in Family Enterprises A. Koeberle-Schmid, D. Kenyon-Rouvinez, E. Poza, INOBUCHI, 2013-12-02 Presents a comprehensive overview of governance in family enterprises including practical management knowledge in easy to use frameworks and interviews with renowned family enterprise owners and managers Readers will benefit from the book's systematic approach and the opportunity to learn from the experience of other family enterprises 1, 2, 3

Code!-Instructor's Manual Ken Green, 2006-08

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**Marketing** William Morgan Pride, O. C. Ferrell, 2010 To maximise this publication's core strengths the authors have included revised concepts features and examples throughout to maintain timely coverage of current marketing trends and strategies

Understanding Health Insurance Brisky, Ruth M. Burke, Michelle A. Green, Jo Ann C. Rowell, Lynette M. Williamson, Jo Ann C. Rowell, 2007-12

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and learning objectives The WiseGuide Wrap Up contains a chapter summary discussion questions and end notes

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