

Inorganic Chemistry

Lesson 1: Inorganic Chemistry

Inorganic Chemistry - deals with **MATTER** its composition, properties and its reactions.

Matter - is anything that occupies space and has mass.

- All the objects in the universe, since they occupy space and have mass, are matter.
- The mass of an object is the quantity of matter that it contains.

Classification of Matter:

1. **PURE SUBSTANCES / SUBSTANCES** have fixed compositions and distinct properties.
 - **ELEMENTS** - are substances that cannot be decomposed into simpler substances by chemical means. They are the basic substances out of which all matter is composed.
 - Metal
 - Nonmetal
 - Metalloid

COMPOUNDS - are the substances composed of two or more elements united chemically in definite proportions by mass. They can be decomposed by chemical means into simpler elements.

- Acid
 - Base
 - Salt
2. **MIXTURES** are composed of two or more kinds of matter where each substance retains its own identity and properties thus it can be separated by physical means. Compositions of mixtures can vary continuously.
- **HOMOGENEOUS MIXTURES (1 phase / SOLUTIONS)** - have uniform composition throughout. Solutions are classified under this group, i.e. *vinegar, brine, syrup.*

→ **HETEROGENEOUS MIXTURES (2/ SUSPENSIONS)** - are substances whose composition, properties, and appearance differ throughout, i.e. sand in water, chocolate chip ice cream, fruit salad.

- ◆ **silverware** - a homogeneous mixture (alloy)

Properties of Matter - refer to the characteristics that enable one to distinguish one substance to another.

1. **Physical properties:** color, odor, taste, hardness, physical state, melting point, boiling point, electrical conductivity, malleability, density, solubility, specific gravity.
2. **Chemical properties:** rusting metal, burning metal, burning fuel, milk turning sour

Physical States Matter

1. **Solid** - the molecules that make up a solid are arranged in regular, repeating patterns. They are held firmly in place but can vibrate within a limited area.
2. **Liquid** - the molecules that make up a liquid flow easily around one another. They are kept from flying apart by attractive forces between them. Liquids assume the shape of their containers.
3. **Gas** - the molecules that make up a gas fly in all directions at great speeds. They are so far apart that attractive forces between them are insignificant.
4. **Plasma** - at the very high temperatures of stars, atoms lose their electrons. The mixture of electrons and nuclei that results is the plasma state of matter.

Changes of Matter

1. **Physical change** does not involve a change of one of matter into another; no change in the fixed composition of the substance.

Inorganic Photochemistry Lecture Notes

R Barnett



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Concepts of Inorganic Photochemistry Arthur W. Adamson, Paul D. Fleischauer, 1975 PHOTOPHYSICAL PROCESSES ENERGY LEVELS AND SPECTRA KINETICS OF PHOTOPHYSICAL PROCESSES CHARGE TRANSFER PHOTOCHEMISTRY SUBSTITUTIONAL PHOTOCHEMISTRY OF FIRST ROW TRANSITION ELEMENTS PHOCHEMISTRY OF THE HEAVIER ELEMENTS PHOTOCHEMISTRY OF CARBONYL COMPLEXES PHOTOCHEMISTRY OF 1 3 DIKETONATE CHELATES THE PHOTOLYSIS OF SIMPLE INORGANIC IONS IN SOLUTION PHOTOCHEMISTRY IN THE SOLID STATE PHOTOCHROMISM AND CHEMILUMINESCENCE *American Scientist* ,1942 *Energy Abstracts for Policy Analysis* ,1975-07 *Directory of Published Proceedings* ,1988 **Inorganic Photochemistry** ,2011-07-14 The Advances in Inorganic Chemistry series present timely and informative summaries of the current progress in a variety of subject areas within inorganic chemistry ranging from bio inorganic to solid state studies This acclaimed serial features reviews written by experts in the field and serves as an indispensable reference to advanced researchers Each volume contains an index and each chapter is fully referenced Features comprehensive reviews on the latest developments Includes contributions from leading experts in the field Serves as an indispensable reference to advanced researchers **Photophysics of Organometallics** Alistair J. Lees, 2009-12-03 Arvind Kumar Shih Sheng Sun and Alistair J Lees Photophysics and Photochemistry of Organometallic Rhenium Diimine Complexes Conor Long Photophysics of CO Loss from Simple Metal Carbonyl Complexes Anton n Vlcek Jr Ultrafast Excited State Processes in Re I Carbonyl Diimine Complexes From Excitation to Photochemistry Kenneth Kam Wing Lo Exploitation of Luminescent Organometallic Rhenium I and Iridium III Complexes in Biological Studies Maria L Muro Aaron A Rachford Xianghuai Wang and Felix N Castellano Platinum II Acetylide Photophysics Andreas F Rausch Herbert H H Homeier and Hartmut Yersin Organometallic Pt II and Ir III Triplet Emitters for OLED Applications and the Role of Spin Orbit Coupling A Study Based on High Resolution Optical Spectroscopy *Pure and Applied Chemistry* ,1970 Some vols include Plenary lectures of the International Conference on Organometallic Chemistry **Emerging Nanomaterials for Energy Conversion and Storage Applications** Guohua Jia, Hongbo Li, Zongyou Yin, 2022-01-04 *Choice* ,2008 **The Cumulative Book Index** ,1999 *Molecular Distortions in Ionic and Excited States* Peter V. Schastnev, Lyudmila N. Shchegoleva, 1995-08-25 This book discusses the achievements in the study on the structure of active species being formed at the change of molecule charge or electronic state It gives a systematic outline of the problem of molecular structure distortions in radical ionic and excited states involving experimental and theoretical material The text focuses on analyzing the physical reasons for structural distortion occurrences at a model level and also with results of detailed quantum chemical calculations The book presents numerous facts on the structural distortions in ions of various types of organic molecules Data is presented for the first time on distortions in radical ions of polyfluoroaromatics and other conjugated molecules **American Book Publishing Record** ,2001 *Cumulated Index to the Books* ,1999 *The Stratosphere and*

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Inorganic Photochemistry Lecture Notes Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Inorganic Photochemistry Lecture Notes**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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