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## AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5.  $q=mc(\Delta T)$  ; What does "q" stand for?: Heat (kJ)
6.  $q=mc(\Delta T)$  ; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12.  $\text{pH} < 7$  ; acid, base, or neutral?: Acid
13.  $\text{pH} > 7$  ; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt,  $\text{pH} = 7$ .
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

# Pre Ap Chemistry Spring Final Review Tenpayore

**G Orfield**



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## Enjoying the Beat of Appearance: An Emotional Symphony within **Pre Ap Chemistry Spring Final Review Tenpayore**

In a global taken by monitors and the ceaseless chatter of fast conversation, the melodic splendor and emotional symphony produced by the written term usually diminish into the backdrop, eclipsed by the persistent sound and distractions that permeate our lives. However, nestled within the pages of **Pre Ap Chemistry Spring Final Review Tenpayore** a wonderful fictional treasure brimming with fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful musician of language, that captivating masterpiece conducts viewers on a mental journey, well unraveling the concealed tunes and profound influence resonating within each cautiously constructed phrase. Within the depths of this poignant review, we shall discover the book is key harmonies, analyze their enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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