

### Atomic Orbitals

Use with Chapter 5,  
Section 5.2

1. What is the shape of an s orbital?  
Spherical
2. What is the relationship between the size of an s orbital and the principal energy level in which it is found?  
\_\_\_\_\_
3. What is the shape of a p orbital? How many p orbitals are there in a sublevel?  
\_\_\_\_\_
4. How many electrons can each orbital hold?  
2
5. Look at the diagrams of the p orbitals. What do x, y, and z refer to?  
\_\_\_\_\_
6. How many d orbitals are there in a given sublevel? How many total electrons can the d orbitals in a sublevel hold?  
5 10
7. Which d orbitals have the same shape?  
 $x^2 - y^2$
8. What point in each diagram represents an atom's nucleus?  
\_\_\_\_\_
9. How likely is it that an electron occupying a p or a d orbital would be found very near an atom's nucleus? What part of the diagram supports your conclusion?  
\_\_\_\_\_

# Teaching Transparency 16 Answers

**Marcel A. Müller**



**Teaching Transparency 16 Answers:**

## Whispering the Strategies of Language: An Emotional Journey through **Teaching Transparency 16 Answers**

In a digitally-driven earth wherever monitors reign supreme and immediate connection drowns out the subtleties of language, the profound strategies and psychological subtleties concealed within phrases frequently move unheard. Yet, nestled within the pages of **Teaching Transparency 16 Answers** a captivating fictional prize blinking with raw thoughts, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, this marvelous opus encourages readers on an introspective journey, lightly unraveling the veiled truths and profound influence resonating within the fabric of every word. Within the emotional depths of the poignant review, we shall embark upon a genuine exploration of the book's primary themes, dissect their charming writing design, and fail to the effective resonance it evokes deep within the recesses of readers' hearts.

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