



### PRE-LAB QUESTIONS

1. What would you estimate the pH of hydrochloric acid, a strong acid, to be? What is the pH of water? The pH of HCl is 3.01, so yes a strong acid, the pH of water is 7
2. Use the Henderson-Hasselbalch equation to estimate the pH of a buffer solution that is composed of 20 mL of 10 M sodium formate and 20 mL of 1 M formic acid (Note: the pKa of formic acid is 3.75). 4.75 would be the pH
3. Look at Table 4 in the procedure portion of the experiment. Calculate the pH you would expect each of the buffer solutions (A, B, C, D, and E) to be using the Henderson-Hasselbalch equation, assuming that the solutions of acetic acid and sodium acetate are equimolar.

# Properties Of Buffer Solutions Pre Lab Answers

**Peipei Pang**



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subscales pdf all tests 2 file of all student data including subscales where applicable csv all tests 3

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web the drumcondra primary spelling test dpst is designed to be administered to pupils in first to sixth classes in primary schools there are two forms of the test at each level form a and form b the dpst takes 45 50 minutes to