

Applying Scientific Methods

A college chemistry student is studying the properties of four unknown compounds, W, X, Y, and Z. She has been informed that one of them is ionic and that the other three are covalent. Of the latter, she has been told that the attractions between the formula units are dispersion forces in one case, hydrogen bonds in another, and covalent (network) bonds in another. She has been asked to determine the type of attractions for each substance. In an attempt to do so, she carries out experiments that reveal information about the substances' properties. Use this information to answer the questions that follow.

Property	Unknown W	Unknown X	Unknown Y	Unknown Z
Hardness of solid	soft	soft	brittle	brittle
Melting point (°C)	-10	-200	1500	3000
Boiling point (°C)	120	-150	2500	4500
Solubility in polar solvent	very soluble	insoluble	very soluble	insoluble
Solubility in nonpolar solvent	insoluble	very soluble	insoluble	insoluble
Conductivity of solid	nonconducting	nonconducting	nonconducting	nonconducting
Conductivity of liquid	nonconducting	nonconducting	conducting	nonconducting

- On the basis of the results of her hardness test, the student concludes that unknowns W and X are covalent substances, and that either Y or Z is the ionic substance. Comment on the soundness of her conclusion, given only the hardness information.

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Table of Contents Chemistry Chapter Assessment Applying Scientific Methods Answers

1. Understanding the eBook Chemistry Chapter Assessment Applying Scientific Methods Answers
 - The Rise of Digital Reading Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter Assessment Applying Scientific Methods Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Personalized Recommendations

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- Chemistry Chapter Assessment Applying Scientific Methods Answers User Reviews and Ratings
- Chemistry Chapter Assessment Applying Scientific Methods Answers and Bestseller Lists
- 5. Accessing Chemistry Chapter Assessment Applying Scientific Methods Answers Free and Paid eBooks
 - Chemistry Chapter Assessment Applying Scientific Methods Answers Public Domain eBooks
 - Chemistry Chapter Assessment Applying Scientific Methods Answers eBook Subscription Services
 - Chemistry Chapter Assessment Applying Scientific Methods Answers Budget-Friendly Options
- 6. Navigating Chemistry Chapter Assessment Applying Scientific Methods Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Chapter Assessment Applying Scientific Methods Answers Compatibility with Devices
 - Chemistry Chapter Assessment Applying Scientific Methods Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Highlighting and Note-Taking Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Interactive Elements Chemistry Chapter Assessment Applying Scientific Methods Answers
- 8. Staying Engaged with Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter Assessment Applying Scientific Methods Answers
- 9. Balancing eBooks and Physical Books Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter Assessment Applying Scientific Methods Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Setting Reading Goals Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter Assessment Applying Scientific Methods Answers
 - Fact-Checking eBook Content of Chemistry Chapter Assessment Applying Scientific Methods Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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