

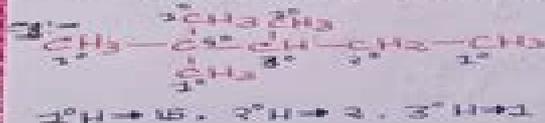
### \* Degree of Carbon :-

- 1° Carbon on. Primary carbon
- 2° Carbon on. Secondary Carbon
- 3° Carbon on. Tertiary Carbon
- 4° Carbon on. Quaternary Carbon



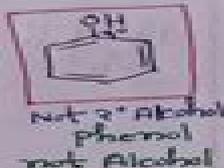
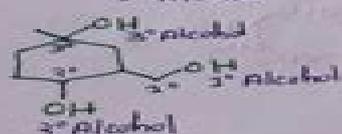
### \* Degree of Hydrogen :-

Same as degree of Carbon.



### \* Degree of Alcohol :-

Same as degree of carbon to which it is attached.



### \* Degree of Halogen :-

Same as degree of carbon to which it attached



1°Cl → 1, 2°F → 2, 3°Br → 3



### \* Degree of Nitrogen/Amine :-

1°/Primary nitrogen/Amine → Nitrogen attached with carbon from one side



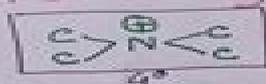
2°/Secondary nitrogen/Amine → Nitrogen attached with carbon from 2 side



3°/Tertiary nitrogen/Amine → Nitrogen attached with carbon from 3 side



4°/Quaternary nitrogen/Amine → Nitrogen attached with carbon from 4 sides and holds one +ve charge



### \* Degree of Unsaturation :- (DU)

Double bond Equivalent / Index of Hydrogen Deficiency

$$DU = \frac{\text{No. of H in corresponding Alkane} - \text{No. of H in given compound}}{2}$$



1 double bond = 2 hydrogen deficiency  
 or. 1 ring = [DU = 1]  
 2 double bond = 4 H- deficiency or [DU = 2]  
 or. 2 ring or. 2 triple bond

# Chemistry S Notes

**BM King**



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