

Pre-lab Questions for Experiment #5

Rates of Chemical Reactions I: A Clock Reaction

1) Complete and balance the following reactions:



2) The rate law for the reaction



is $\text{rate} = k[H_2][NO]^2$. Which of the following mechanisms can be ruled out on the basis of the observed rate expression?

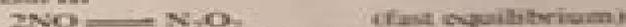
Mechanism I



Mechanism II



Mechanism III



3) Consider the reaction



From the following data obtained at a certain temperature, determine the order of the reaction and calculate the rate constant:

[A] (M)	[B] (M)	Rate (M/s)
1.50	1.50	3.20×10^{-1}
1.50	2.50	3.20×10^{-1}
3.00	1.50	6.40×10^{-1}

4) The following gas-phase reaction was studied at 290°C by observing the change in pressure as a function of time in a constant-volume vessel:



Determine the order of the reaction and the rate constant based on the following data:

Time (s)	Partial Pressure of $ClCO_2CCl_3$ (mmHg)
0	15.76
181	12.64
513	8.73
1164	4.44

Chem Fax Lab 16 Answers

**Theodore Silver, Princeton Review
(Firm)**



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The Chemical News and Journal of Industrial Science ,1919 *American Medical Directory* ,1912 **Question and Answer Review for Phlebotomy** Kathleen Becan-McBride,Diana Garza,2000-11 Prentice Hall Health Review Series About the book Q A Review for Phlebotomy by Drs Kathleen Becan McBride and Diana Garza has helped thousands of students pass their certification exams throughout the years Now in its landmark fifth edition the book blends its comprehensive collection of practice exam questions with the exciting possibilities of computer technology As you build confidence by digging into this rich content review you ll find that the Prentice Hall Health test preparation system is a blueprint for success across the boards Boost Your Scores Launch Your Career Our technology solutions allow you to gain more practice and review Log onto www.prenhall.com review or load the CD ROM that accompanies this book to find additional questions rationales and links to related resources The CD ROM includes an audio glossary and over 750 exam style questions About the Authors Kathleen Becan McBride EdD MT ASCP CLS NCA is the Director of Community Outreach Education at the University of Texas Health Science Center at Houston She has published 13 books and over 50 journal articles She has been educating students in the health professions since 1973 Diana Garza EdD MT ASCP CLS NCA is Associate Research Professor and Project Director for Interdisciplinary Health Care Education at Texas Woman s University Institute of Health Sciences in Houston Texas She is also an Adjunct Clinical Associate Professor in the Department of Laboratory Medicine at the University of Texas M D Anderson Cancer Center in Houston She has published nine books numerous journal articles and has been educating students in the health professions since1981 About the Series Success Across the Boards is a study system that revolutionizes exam preparation for a variety of health care professions By integrating interactive technology with authoritative content and realistic sample test questions Q A Review for Phlebotomy will help you join the many others who have used these materials to achieve success on the boards The series is comprised of test preparation resources for the following health care professions Dental Hygiene Medical Assisting Nursing Phlebotomy Emergency Medical Services *Reports from Commissioners* Great Britain. Parliament. House of Commons,1875

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Silver, Princeton Review (Firm), 2001-03-15 The Princeton Review realizes that acing the SAT II Chemistry exam is very different from getting straight As in school They don't try to teach students everything there is to know about chemistry only what they'll need to score higher on the exam There's a big difference In *Cracking the SAT II Chemistry* The Princeton Review will teach test takers how to think like the test makers and Learn test taking strategies that will help students outsmart the test and improve scores Ace the exam by becoming familiar with the format Use the Process of Elimination and the divide and conquer method to solve complicated problems Perfect test taking skills with practice questions and detailed answer explanations This book includes 2 full length simulated SAT II Chemistry exams All of the sample test questions are just like the ones test takers will see on the actual exam and every solution is fully explained Contents Include I Introduction II Test Strategies III Some Basic Stuff Mass Volume Density Pressure Energy Temperature and Specific Heat IV Elements Atoms and Ions Atoms and Elements V Chemical Reaction and Stoichiometry Molecules The Mole Chemical Reactions Reaction Stoichiometry Entropy Enthalpy Spontaneity and Gibbs Free Energy VI Electron configurations and Radioactivity Electrons and Orbitals Radioactivity VII The Periodic Table and Bonding The Periodic Table More About the Periodic Table Some Important Trends VIII Solids Liquids and Gases Gases Intermolecular Forces Phase Changes Energy and Phase Changes IX Solutions Solutions Concentrations Solubility and Saturation X Kinetics and Equilibrium Kinetics Factors that Affect Reaction Rate Reversible Reactions and Chemical Equilibrium Le Chatelier's Principle XI Acids and Bases Acids and Bases Titration XII Redox and Electrochemistry Oxidation and Reduction Electrochemistry XIII Organic Chemistry Hydrocarbons Functional Groups XIV Laboratory Safety Rules Accuracy Significant Figures Lab Procedures Laboratory Equipment XV Practice Tests

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American Horticultural Manual: Comprising the leading principles and practices connected with the propagation, culture, and improvement of fruits, nuts, ornamental trees, shrubs, and plants in the United States and Canada Joseph L. Budd, Niels Ebbesen Hansen, 1902 **Outlines of Theoretical Chemistry** Frederick Hutton Getman, 1913 **The English Catalogue of Books** Sampson Low, 1916 Volumes for 1898 1968 include a directory of publishers

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