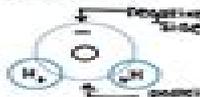


Structure of H₂O
hydrogen bonding
Dimer of H₂O



unit one guide

Biological molecules
proteins
nucleic acids

H₂O provides cushion, adhesion, surface tension, high specific heat, high heat of vaporization, excellent solvent, pH buffer, temperature stability

hydrogen bonds require lots of energy to change state of matter

CONDENSATION

when H₂O bonds in products hydrogen and hydroxyl ions, releases pH → more hydrogen

Dehydration → attracted to same
Hydration → attracted to other
Hydrolysis → breaking down
Hydrolysis → breaking down
Hydrolysis → breaking down

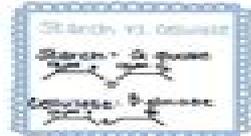
| | | | | | |
|----------|----------|----------|----------|------------|----------|
| C | H | N | O | P | S |
| Carbon | Hydrogen | Nitrogen | Oxygen | Phosphorus | Sulfur |

more hydrogen bonds → C⁻ hold together molecules, cell formation
N⁺ hold positive, amino acids
P⁻ hold nucleic acid + lipids
→ make up 98% of organisms



Dehydration synthesis is the removal of an H₂O molecule to bond monomers
Hydrolysis is the addition of a water molecule to break a bond between monomers (amino acids, nucleic acids, lipids)

Nucleic acid structure:
* the carbon sugar (deoxyribose/ribose), phosphate, nitrogen base (A, T, C, G), exact shape of strand



Proteins: specific order of amino acids

Primary structure → specific sequence of amino acids



amino acids bond via a peptide bond (covalent) through dehydration synthesis
R group change can cause a hydrophobic protein to lose the place of a hydrophilic protein

membranes

made up of phospholipids
create selectively permeable barrier
contain phospholipids, cholesterol, and proteins



LIPIDS

Phospholipid macromolecules
differences in saturation determine structure and function
Cells, sterols, phospholipids
Saturated → no double bonds
Unsaturated → has double bonds
4 in 1: triglycerides, cholesterol, make up the membrane



Amino Acids: have directionality

linear chains, connected by covalent bonds
amino end and carboxyl end

DEHYDRATION SYNTHESIS



Chaperones:

protein molecules that assist protein folding

PROTEIN STRUCTURE

- Primary:** sequence of amino acids, specific order determine protein shape
- Secondary:** how folding of chain into alpha helix and beta sheet occurs
- Tertiary:** 3-D shape, hydrogen bonds, disulfide link, ionic bonds, hydrophobic interactions
- Quaternary:** interactions between multiple polypeptide chains

+ FUNCTION

- Enzymes → speeds up reaction rates in the cell
- Cell structure → part of the cellular membrane
- Communication between cells

DENATURATION:

- the protein starts to lose its shape/structure when (bonds break)
 - temperature change
 - pH of surrounding area
 - salt concentration

NH₂ = Amino
COOH = Carboxyl

NUCLEIC ACIDS

- add new nucleotides to the 3' end
- two nitrogenous bases: purine + pyrimidine
purine is single ring, pyrimidine is double
- antiparallel double helix
- 5' on phosphate end, 3' on hydroxyl end
- both back and front have nitrogenous bases (phospholipids) → sugar-phosphate backbone

ANTIPARALLEL



ALDOSE

contains aldehyde functional group

VS KETOSE

contains ketone functional group



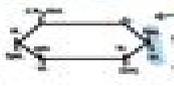
ALPHA VS BETA

all monomers are in the same orientation



BETA

every glucose monomer is upside down from its neighbor



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