

MODULE - 9 LECTURE NOTES - 4

DECISION SUPPORT SYSTEMS

INTRODUCTION

DSS is an interactive computer-based system to help decision makers use communications technologies, data, documents, knowledge and/or models to identify and solve problems, complete decision process tasks, and make decisions. Geographic Information Systems (GIS), Enterprise Information Systems (EIS), Expert Systems (ES), On-Line Analytical Processing (OLAP), software agents, knowledge discovery systems and group DSS can all be lumped into the category of systems we call as DSS.

DSS FOR DECISION-MAKING

Decision Preparation

- DSS environments provide data required as input to the decision-making process.
- This is what almost all data mart and data warehousing environments do today.

Decision structuring

- DSS environments provide tools and models for arranging the inputs in ways that make sense to frame the decision.
- These tools and models are not pivot tables and other aspects of data presentation found in query tools.
- They are actual decision-making tools, like fault tree analysis, Bayesian logic and model-based decision-making based on things like neural networks.

Context Development

- DSS environments again provide tools, and provide the mechanisms for capturing information about a decision's constituencies (who are affected by this decision), outcomes and their probabilities, and other elements of the larger decision-making context.

Decision-making

- DSS environments may automate all or part of the decision-making process and offer evaluations on the optimal decision.
- Expert systems and artificial intelligence environments can do this

Decision Propagation

- DSS environments take the information gathered about constituencies, dependencies, outcomes and drive elements of the decision into those constituencies for action.

Module 9 Lecture Notes Decision Support Systems

Introduction

VM Jensen



Module 9 Lecture Notes Decision Support Systems Introduction :

Embracing the Tune of Expression: An Psychological Symphony within **Module 9 Lecture Notes Decision Support Systems Introduction**

In some sort of eaten by screens and the ceaseless chatter of quick connection, the melodic elegance and psychological symphony produced by the written term frequently disappear into the background, eclipsed by the persistent noise and interruptions that permeate our lives. But, set within the pages of **Module 9 Lecture Notes Decision Support Systems Introduction** a marvelous fictional value brimming with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this fascinating masterpiece conducts readers on a psychological journey, skillfully unraveling the hidden melodies and profound influence resonating within each carefully constructed phrase. Within the depths of the touching assessment, we can examine the book is main harmonies, analyze its enthralling writing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-} + \text{Fe}^{2+} + \text{Cr}^{3+} + \text{Fe}^{3+}$ solution 1 balanced half reactions $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+} + \text{e}^-$ 2 equalize the electrons $6\text{e}^- + 14\text{H}^+ + \text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$ $6\text{Fe}^{2+} \rightarrow 6\text{Fe}^{3+} + 6\text{e}^-$ multiplied by a factor of 6

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