

Physics Notes Class 12 Chapter 14

Semiconductor Electronics, Materials, Devices and Sample Circuits

It is the branch of science which deals with the electron flow through a vacuum, gas or semiconductor.

Classification of substances on the basis of conduction of electricity.

Solid

We know that, each substance is composed of atoms. Substances are mainly classified into three categories namely solids, liquids and gases.

In each solid atoms are at a definite positions and the average distance between them is constant.

Depending upon the internal arrangement of atoms, solids are further divided into two groups.

1. Crystalline Solids

The solid in which the atoms are arranged in a regular order are called the crystalline solids. In other words, we can say that in a crystalline solid, there is periodicity and regularity of its component atoms in all the directions. For example sodium chloride (common salt), diamond, Sugar, silver etc are the crystalline solids.

Their atoms are arranged in a definite geometrical shape.

They have a definite melting point.

They are anisotropic, i.e., their physical properties such as thermal Conductivity refractive index etc, are different in different directions.

They are the real solids.

2. Amorphous Solids

The Solids in which the atoms do not have a definite arrangement are called the amorphous solids. They are also called the glassy solids. For example glass, rubber, plastic, power, etc are the amorphous solids.

They do not have a definite arrangement of its atoms, i.e., they do not have a characteristic geometrical shape.

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Table of Contents Physics Notes Class 12 Chapter 14 Semiconductor

1. Understanding the eBook Physics Notes Class 12 Chapter 14 Semiconductor
 - The Rise of Digital Reading Physics Notes Class 12 Chapter 14 Semiconductor
 - Advantages of eBooks Over Traditional Books
2. Identifying Physics Notes Class 12 Chapter 14 Semiconductor
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics Notes Class 12 Chapter 14 Semiconductor
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Notes Class 12 Chapter 14 Semiconductor
 - Personalized Recommendations
 - Physics Notes Class 12 Chapter 14 Semiconductor User Reviews and Ratings
 - Physics Notes Class 12 Chapter 14 Semiconductor and Bestseller Lists
5. Accessing Physics Notes Class 12 Chapter 14 Semiconductor Free and Paid eBooks
 - Physics Notes Class 12 Chapter 14 Semiconductor Public Domain eBooks
 - Physics Notes Class 12 Chapter 14 Semiconductor eBook Subscription Services
 - Physics Notes Class 12 Chapter 14 Semiconductor Budget-Friendly Options
6. Navigating Physics Notes Class 12 Chapter 14 Semiconductor eBook Formats

- ePub, PDF, MOBI, and More
 - Physics Notes Class 12 Chapter 14 Semiconductor Compatibility with Devices
 - Physics Notes Class 12 Chapter 14 Semiconductor Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics Notes Class 12 Chapter 14 Semiconductor
 - Highlighting and Note-Taking Physics Notes Class 12 Chapter 14 Semiconductor
 - Interactive Elements Physics Notes Class 12 Chapter 14 Semiconductor
 8. Staying Engaged with Physics Notes Class 12 Chapter 14 Semiconductor
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics Notes Class 12 Chapter 14 Semiconductor
 9. Balancing eBooks and Physical Books Physics Notes Class 12 Chapter 14 Semiconductor
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics Notes Class 12 Chapter 14 Semiconductor
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physics Notes Class 12 Chapter 14 Semiconductor
 - Setting Reading Goals Physics Notes Class 12 Chapter 14 Semiconductor
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physics Notes Class 12 Chapter 14 Semiconductor
 - Fact-Checking eBook Content of Physics Notes Class 12 Chapter 14 Semiconductor
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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