

## Unit 1: Atomic Structure

(Also located on \_\_\_\_\_)

D. Ways to describe atomic info:

1. \_\_\_\_\_  
" \_\_\_\_\_

**Example:** An atom has an atomic number of 8. How many protons are present in this atom?

1. \_\_\_\_\_  
" \_\_\_\_\_  
" \_\_\_\_\_

**You Try:**

Atoms of element Hydrogen have: \_\_\_\_\_

Atoms of element Oxygen have: \_\_\_\_\_

Atoms of element Iron have: \_\_\_\_\_

1. \_\_\_\_\_  
" \_\_\_\_\_  
" \_\_\_\_\_

**Example:** An atom has 6 protons and 8 neutrons. What is this atom's atomic mass number?

If an atom has a mass number of 20 and an atomic number of 8, how many neutrons are present in this atom?

# Atomic Structure Guided Notes Answers Boypic

**L Reisser**



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