

Atomic Structure and Interatomic Bonding

(Chapter 2 of Callister)

Coordinator: An Mai, PhD, PEng



What is an Atom?

A HISTORY OF THE ATOM: THEORIES AND MODELS

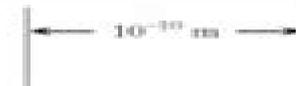
How have our ideas about atoms changed over the years? This graphic looks at atomic models and how they developed.

SOLID SPHERE MODEL	PLUM PUDDING MODEL	NUCLEAR MODEL	PLANETARY MODEL	QUANTUM MODEL
JOHN DALTON	J.J. THOMSON	ERNEST RUTHERFORD	NIELS BOHR	ERWIN SCHRÖDINGER
1803	1904	1911	1913	1926
<p>Atoms are made of tiny particles called molecules. These molecules are made of atoms. Atoms are the smallest particles that cannot be created or destroyed. Atoms are the building blocks of matter.</p> <p>Atoms are made of three subatomic particles: electrons, protons, and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.</p> <p>Atoms are electrically neutral. The number of protons is equal to the number of electrons.</p>	<p>Atoms are made of three subatomic particles: electrons, protons, and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.</p> <p>Atoms are electrically neutral. The number of protons is equal to the number of electrons.</p> <p>Electrons are embedded in a positively charged matrix called the plum pudding.</p>	<p>Atoms are made of three subatomic particles: electrons, protons, and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.</p> <p>Atoms are electrically neutral. The number of protons is equal to the number of electrons.</p> <p>Electrons revolve around the nucleus in elliptical orbits.</p>	<p>Atoms are made of three subatomic particles: electrons, protons, and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.</p> <p>Atoms are electrically neutral. The number of protons is equal to the number of electrons.</p> <p>Electrons revolve around the nucleus in concentric circular orbits.</p>	<p>Atoms are made of three subatomic particles: electrons, protons, and neutrons. Electrons are negatively charged, protons are positively charged, and neutrons are neutral.</p> <p>Atoms are electrically neutral. The number of protons is equal to the number of electrons.</p> <p>Electrons are present in a diffuse cloud around the nucleus.</p>
<p>✓ REPRESENTATION OF ATOMS</p> <p>✓ ATOMS ARE MADE OF TINY PARTICLES CALLED MOLECULES.</p>	<p>✓ REPRESENTATION OF ATOMS</p> <p>✓ ATOMS ARE MADE OF TINY PARTICLES CALLED MOLECULES.</p>	<p>✓ REPRESENTATION OF ATOMS</p> <p>✓ ATOMS ARE MADE OF TINY PARTICLES CALLED MOLECULES.</p>	<p>✓ REPRESENTATION OF ATOMS</p> <p>✓ ATOMS ARE MADE OF TINY PARTICLES CALLED MOLECULES.</p>	<p>✓ REPRESENTATION OF ATOMS</p> <p>✓ ATOMS ARE MADE OF TINY PARTICLES CALLED MOLECULES.</p>
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Atoms, Molecules, and Compounds



Chapter 2 Atomic Structure Interatomic Bonding And

C Cleary



Chapter 2 Atomic Structure Interatomic Bonding And:

The book delves into Chapter 2 Atomic Structure Interatomic Bonding And. Chapter 2 Atomic Structure Interatomic Bonding And is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chapter 2 Atomic Structure Interatomic Bonding And, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 2: Essential Elements of Chapter 2 Atomic Structure Interatomic Bonding And
 - Chapter 3: Chapter 2 Atomic Structure Interatomic Bonding And in Everyday Life
 - Chapter 4: Chapter 2 Atomic Structure Interatomic Bonding And in Specific Contexts
 - Chapter 5: Conclusion
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