

Name _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____.
A) the solvent is a gas and the solute is a solid
B) they are accompanied by an increase in order
C) they are accompanied by an increase in disorder
D) they are accompanied by another process that is exothermic
E) the solvent is water and the solute is a gas
- 2) The phrase "like dissolves like" refers to the fact that _____.
A) polar solvents dissolve nonpolar solutes and vice versa
B) solvents can only dissolve solutes of similar molar mass
C) gases can only dissolve other gases
D) polar solvents dissolve polar solutes and nonpolar solvents dissolve nonpolar solutes
E) condensed phases can only dissolve other condensed phases
- 3) An unsaturated solution is one that _____.
A) has no double bonds
B) has a concentration lower than the solubility
C) contains no solute
D) contains the maximum concentration of solute possible, and is in equilibrium with undissolved solute
E) contains more dissolved solute than the solubility allows
- 4) A solution with a concentration higher than the solubility is _____.
A) is not possible
B) is saturated
C) is supercritical
D) is unsaturated
E) is supersaturated
- 5) Which one of the following substances would be the most soluble in CCl_4 ?
A) H_2O B) NaCl C) $\text{CH}_3\text{CH}_2\text{OH}$ D) NH_3 E) $\text{C}_{10}\text{H}_{22}$
- 6) Which one of the following is most soluble in water?
A) CH_3OH
B) $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$
C) $\text{CH}_3\text{CH}_2\text{OH}$
D) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$
E) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

Chapter 11 Chemistry Test

Mary Louise Turgeon



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