

## Chapter 12

### SECTION 12.1

#### 12.1.1 Boyle's Law: Pressure-Volume Relationship for an Ideal Gas

1. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume. Illustrate Boyle's Law with the following experimental setup:

Boyle's Law: A cylindrical container with a piston is partially filled with gas.

Boyle's Law: The piston is pushed down, compressing the gas.

Boyle's Law: The piston is pulled up, expanding the gas.

Boyle's Law: The piston is pushed down, compressing the gas.

2. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume. Illustrate Boyle's Law with the following experimental setup:

Boyle's Law: A cylindrical container with a piston is partially filled with gas.

Boyle's Law: The piston is pushed down, compressing the gas.

Boyle's Law: The piston is pulled up, expanding the gas.

Boyle's Law: The piston is pushed down, compressing the gas.

3. Boyle's Law states that pressure of a fixed mass of gas is inversely proportional to its volume. Illustrate Boyle's Law with the following experimental setup:

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# Chapter 11 Review Gases Answer Key

**Daniel F McAuley**



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